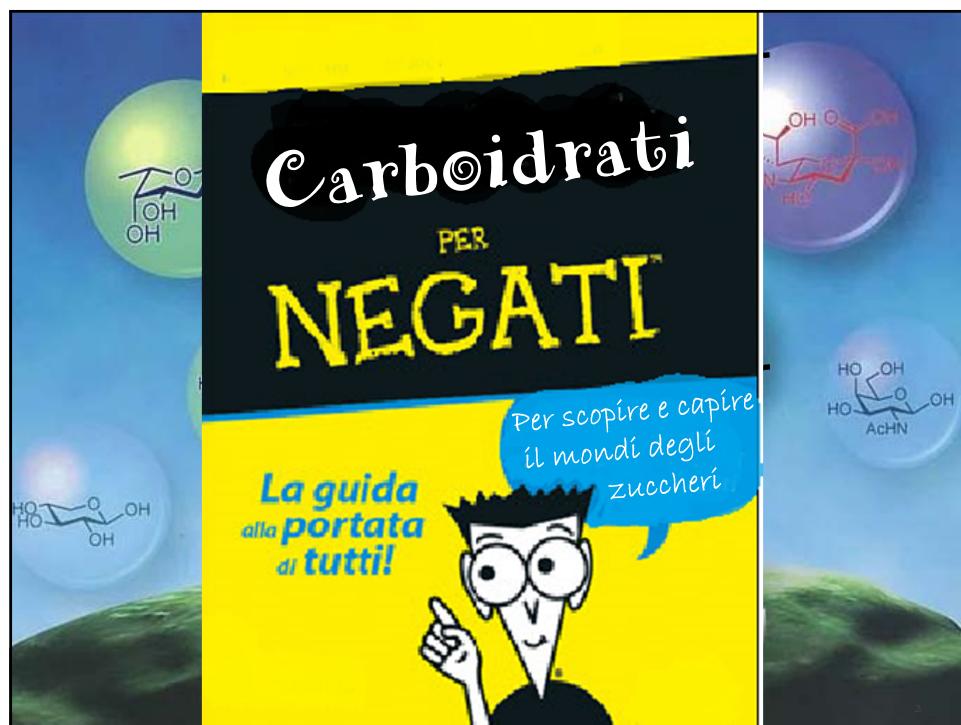


1



1

Sucrose

Sugar cane : was discovered in New Guinea, arrived to India around 600 BC and reached Persia when Darius invaded India. Upon invading Spain, Arabs brought sugar to Europe around 700 AD.

Etymology : sanskrit "SARKARA" (grain), translated as *sukkar* in arabic, *saccharum* in latin and *zucchero* in italian

Western Europe : average consumption 25 kg of sucrose par year per capita

Sugar beat : identified by a German Chemist, Andreas Marggraf en 1747; and brought to culture under Napoleon on the occasion of the british blockade.

Production of 145 millions tons/year of a « molecularly » pure product !

Chemical Structure:

Disaccharide
D-glucopyranose
D-fructofuranose

O=C1OC(O)C(O)C(O)C(O)C1O

What do we mean by « Sugars » ?

Refer to a vast family of molecules, which are **homogeneous** ($C_x (H_2O)_y$) and **diversified**.

Made up of **Carbone, Hydrogen & Oxygen**.

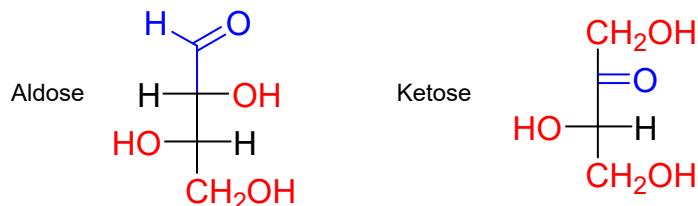
They are referred to as « **Carbohydrates** », « **Hydrates de carbone** » « **Kohlenhydrate** » but also « **Saccharides** » and « **Glycans** »

Oligosaccharides, Polysaccharides

Glycoconjugates as in Glycopeptides, Glycoproteins, Glycolipids

Molecular Diversity of Monosaccharides

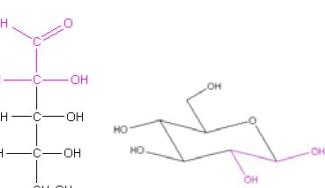
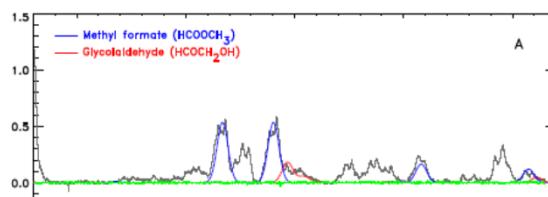
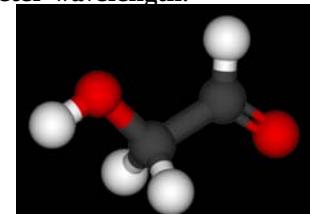
- Carbon chain (3 à 10 Carbons) multiforms
- All Carbons are functionalized
- Coexistence of a **Carbonyl group** (aldehyde or ketone) with several (2 to 9) **Hydroxyl groups** : aldoses / ketoses
- Dual character **Electrophile – Nucleophile**



Sugar in Space

(400 light-years away to Earth)

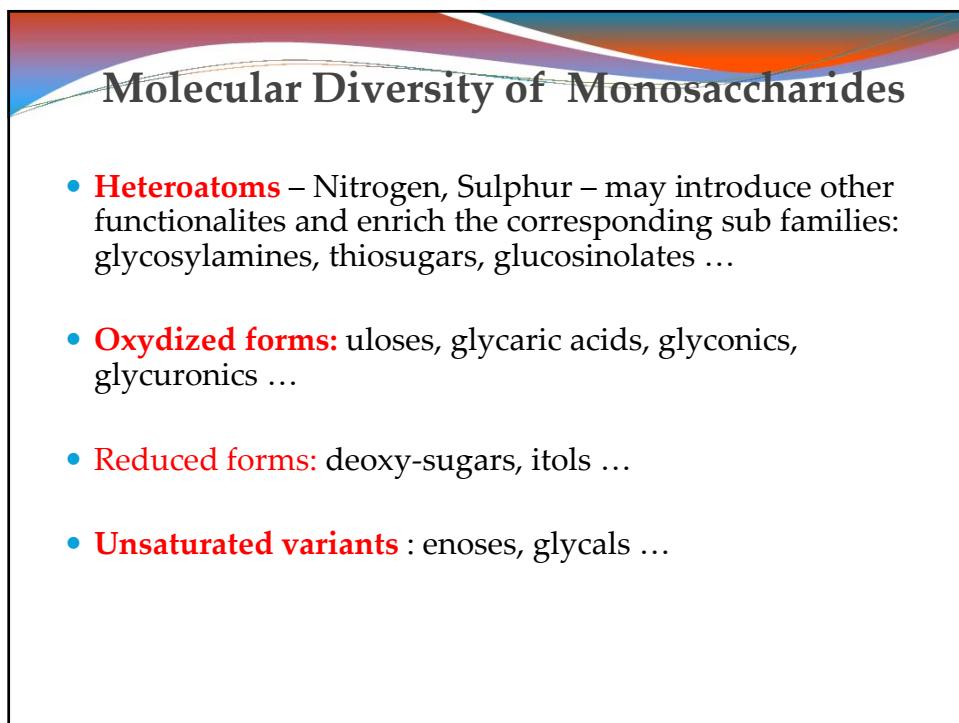
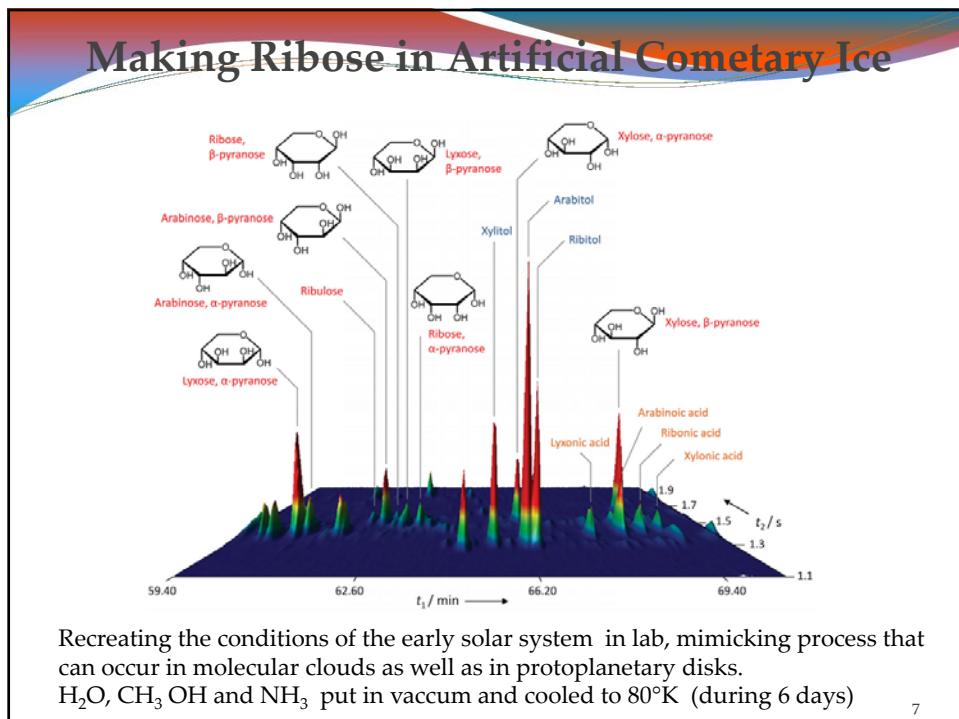
Glycoaldehyde (and its reduced alcohol variant – ethylene glycol) have been identified from the energy it emits as it changes its rotational energy level by absorbing and emitting in the millimeter wavelength.



Ribose → ARN → ADN

Jørgensen, J. K.; Favre, C.; Bisschop, S.; Bourke, T.; Dishoeck, E.; Schmalzl, M. (2012). Detection of the simplest sugar, glycoaldehyde, in a solar-type protostar with ALMA (http://www.eso.org/public/archives/science_papers/eso1234/eso1234a.pdf)

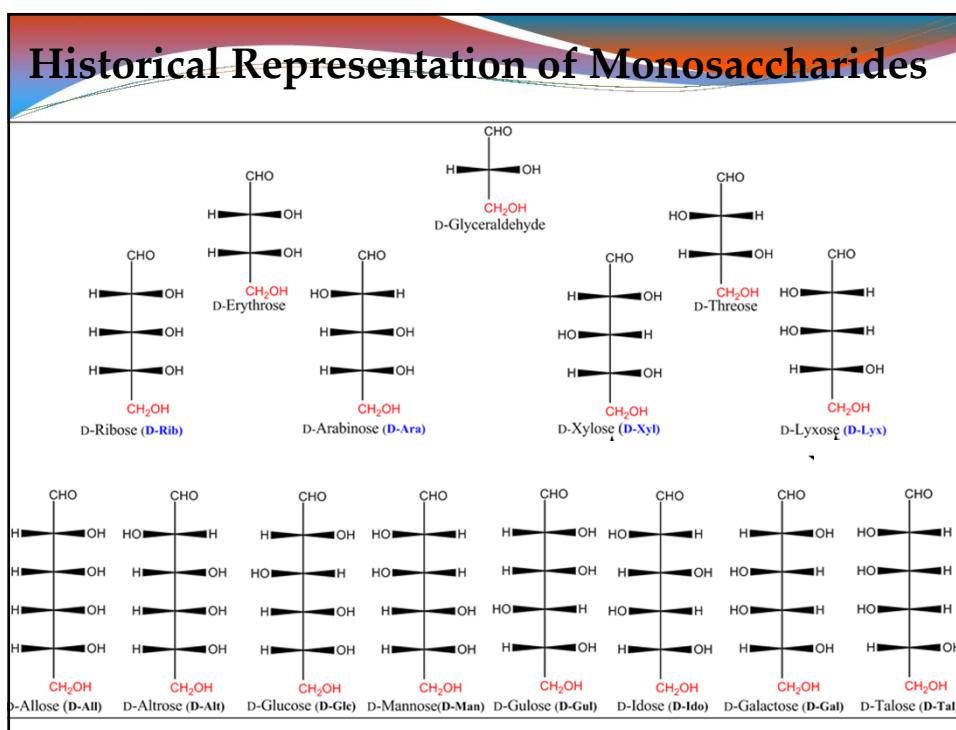
6



Chirality & Stereochemistry

CHO
H — OH
HO — H
H — OH
H — OH
CH₂OH

9

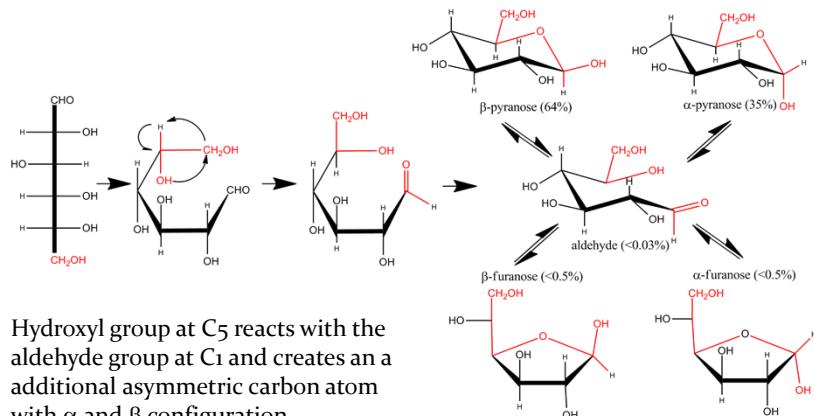


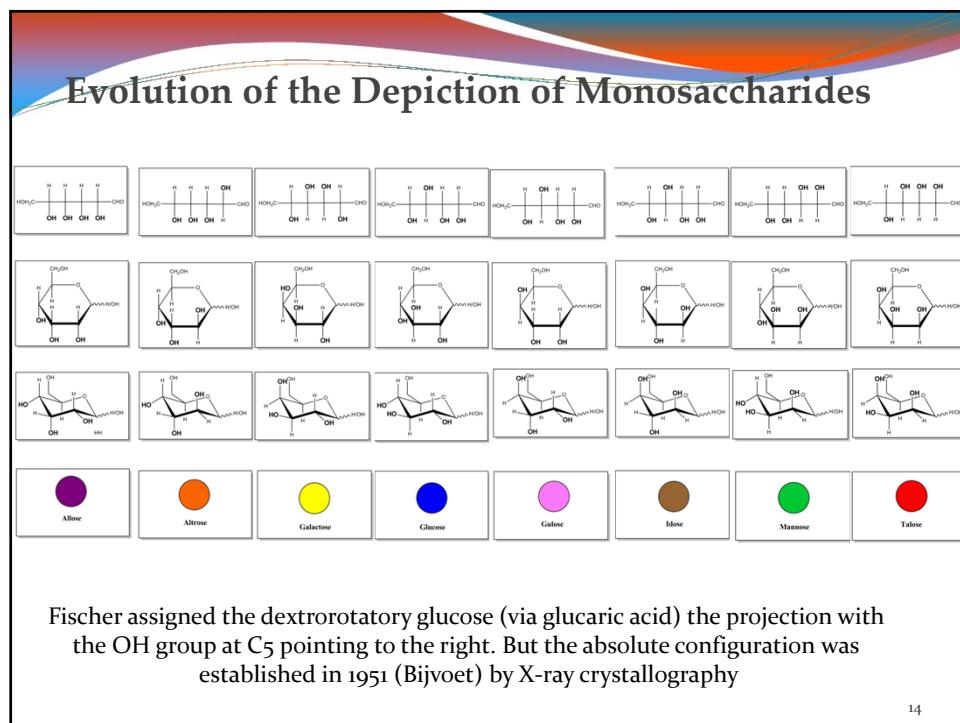
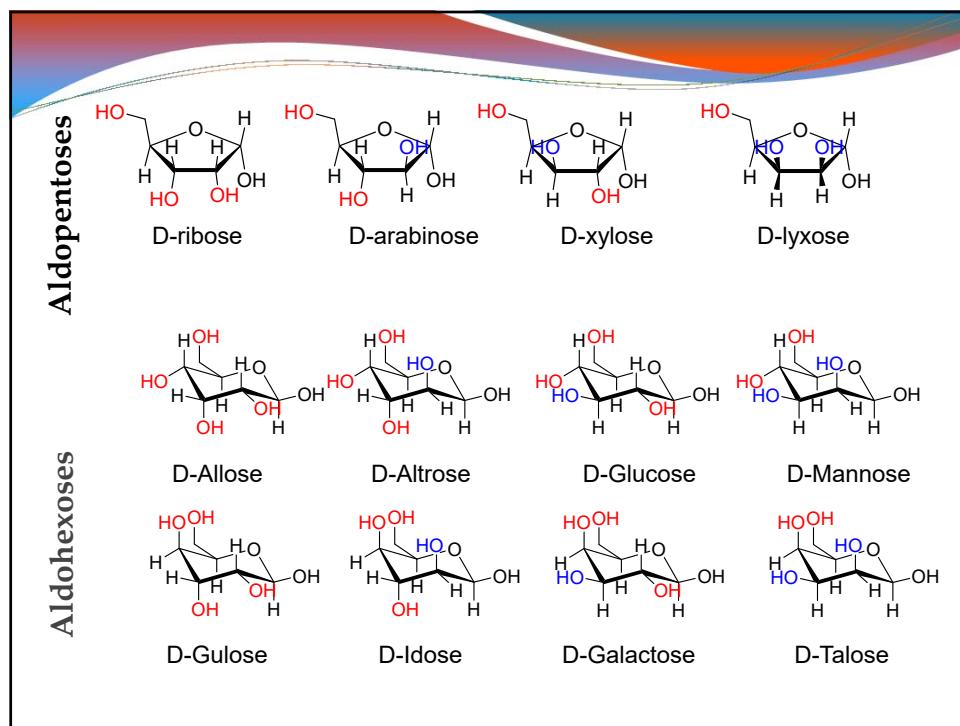
Equilibrium Chain - Ring

- There exists an **equilibrium** (reversible or not) between the linear chains and the cyclic form.
Its formation is favored under weak acidic or alkaline conditions).
- The more stable rings are (4C & 1 O) (**furanose**) and (5C & 1O) (**pyranose**).
- Ring locking generates a crucial supplementary complexity: ***the anomeric carbon: with two configurations.***

either α , or β

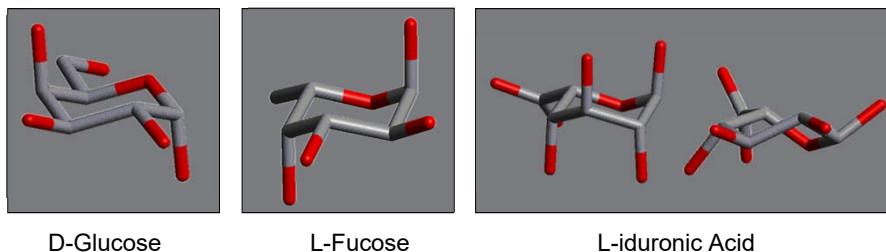
Mutarotation : Configurational Equilibrium





Conformational Complexity

- The most common rings are furanoid or pyranoid, with five and six members.
- Furanose ring structures occur in Envelope (E) and Twist (T) conformations.
- Six membered ring structures can occur in 2 chairs (C), 6 boat (B), 6 Skew (S) and 12 half-chair (H) conformations.
- These conformations result in orientating in space the different functions of the monosaccharide, which have an impact on recognition and reactivity.



The Isomer Barrier

Nucleotides	Peptides	Glycans
4 bases	20 amino acids	> 150 (700) monosaccharides
Linear	Linear	Lineare or branchéed
2 sites linkage	4 sites linkage	5 sites linkage With 2 stéréo-chemistry (α / β)
Hexanucleotide $4^6 = 4096$	Hexapeptide $20^6 = 64\ 000\ 000$	Hexasaccharide 192 780 943 360 (with 10 monosaccharides)

High Coding Capacity of Glycans

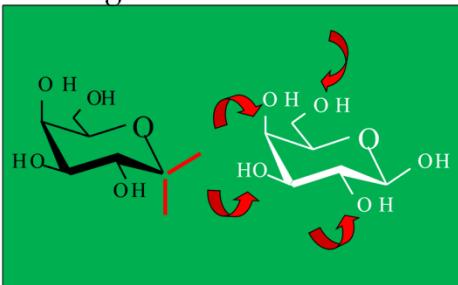
R. A. Laine, Glycobiology, 4, 759-767 (1994)

P. H. Seeberger, ACS Chem Biol, 2, 685-691 (2007)

16

The Isomer Barrier

- Have a very high number of monomers (substitution...).
- Have many different ways of connecting monomers.
- Have branching points.
- Are difficult to synthesise ... and to crystallize.
- Are not the direct product of a gene (\neq proteins).
- Cannot be amplified by PCR (\neq Nucleic acids).



17

A New and Complex Alphabet

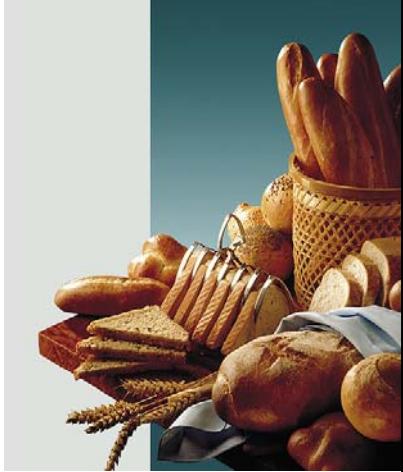
English <chem>...TTGCCTAGTGATTGGATG...</chem> <chem>...LysProSerAspTrpMet...</chem>	東 東  "Hemoglobin"
	"I am a cancer cell"

BioInformatics: Difficulty to encode the structures

18

Oligosaccharides & Polysaccharides

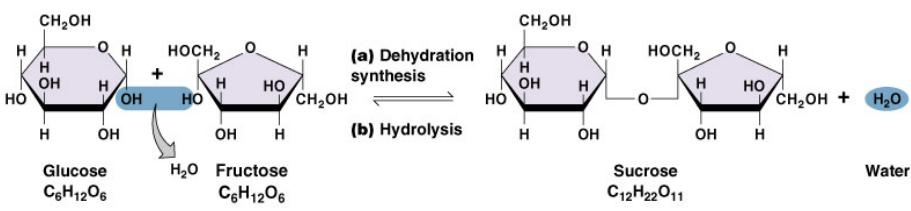
**Structural Complexity
and
Functional Diversity**



Oligo & Polysaccharides :

Monosaccharide units linked by glycosidic linkage

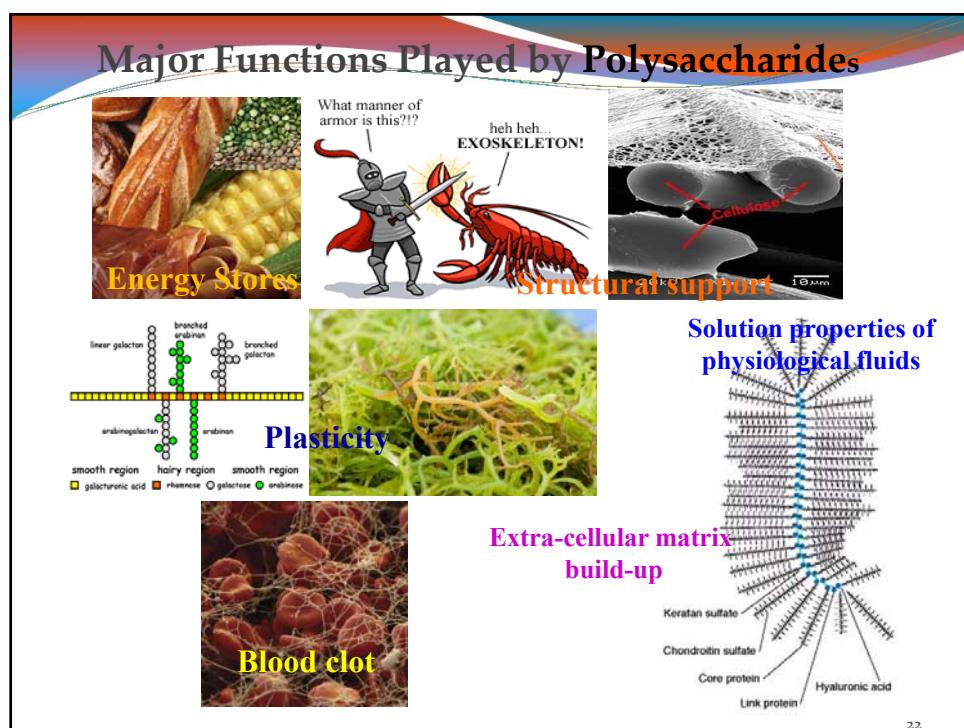
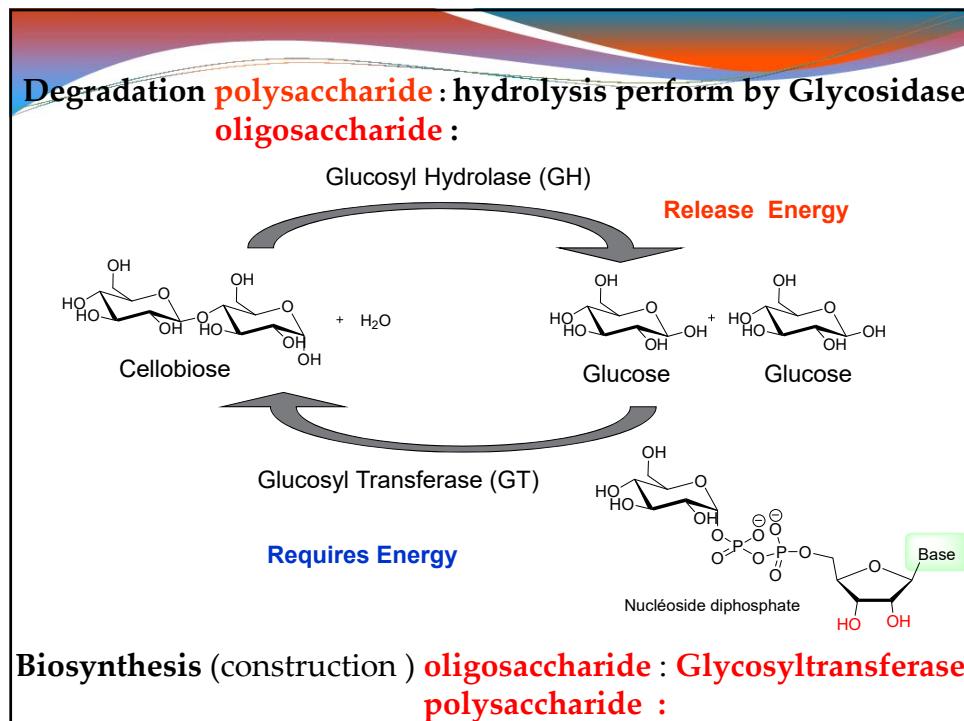
One of the most stable linkage among those occurring in bio(macro)molecules

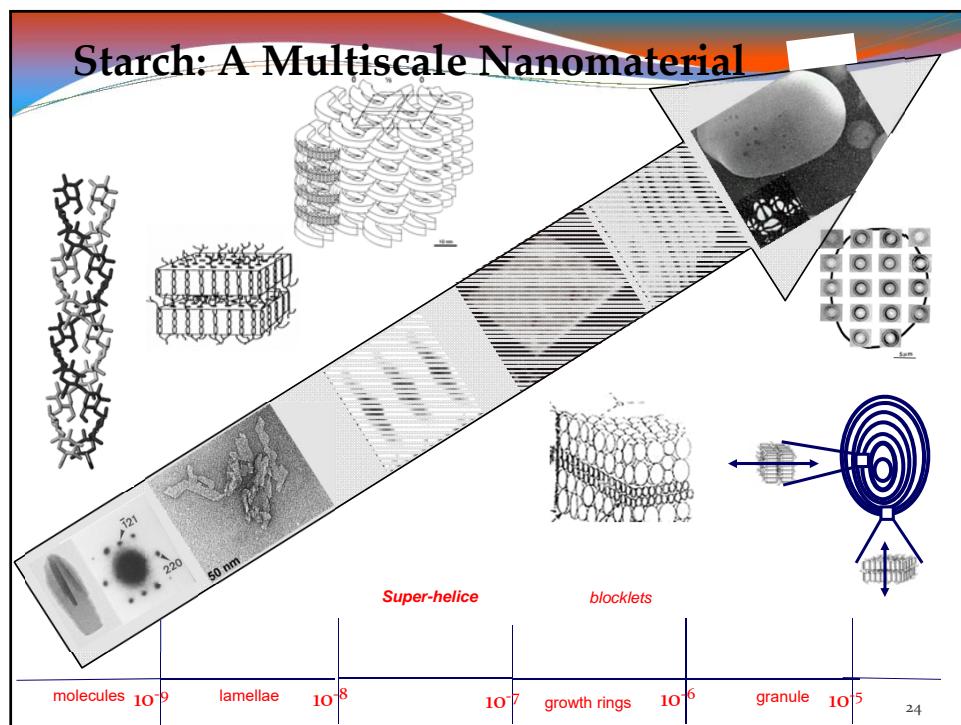
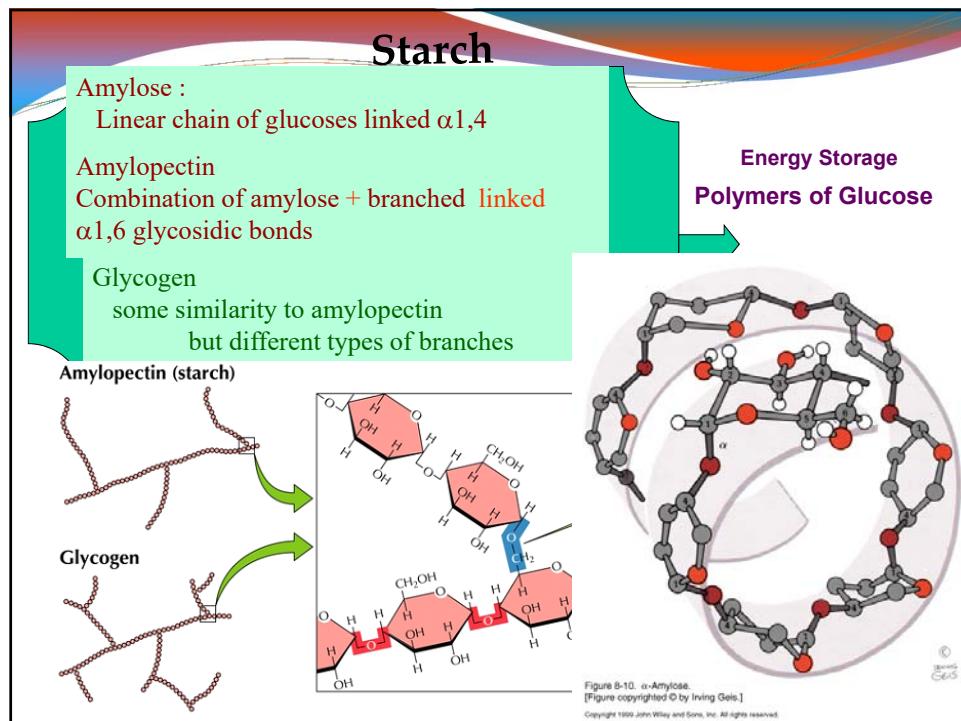


The diagram illustrates the reversible condensation synthesis of sucrose from glucose and fructose, and its hydrolysis back into glucose and fructose with the release of water.

(a) Dehydration synthesis: Glucose ($C_6H_{12}O_6$) and Fructose ($C_6H_{12}O_6$) react with the loss of water (H_2O) to form Sucrose ($C_{12}H_{22}O_{11}$) and a glycosidic linkage.

(b) Hydrolysis: Sucrose ($C_{12}H_{22}O_{11}$) reacts with water (H_2O) to regenerate Glucose ($C_6H_{12}O_6$) and Fructose ($C_6H_{12}O_6$).



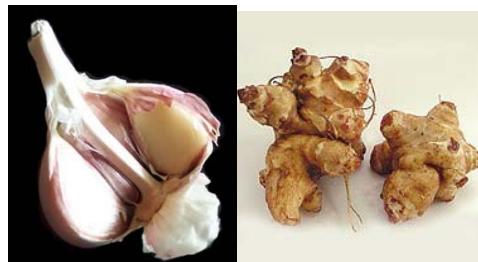


Other Energy Storage Polysaccharides,

(with linear structures more rare, found in plants and small animals)

galactans (in snails) ; 20-30 units

fructane (plants) 20-30 units



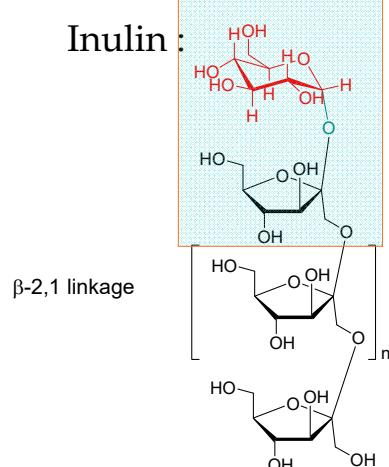
Helianthus tuberosus

Jerusalem artichoke
Topinambour
Topinanbur
(carciofo di Gerusalemme)

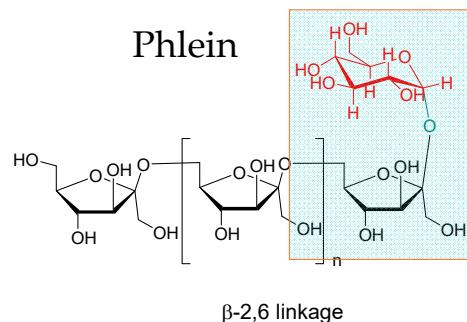
Fructans : Two Types of Structures :

Inulin Phlein types

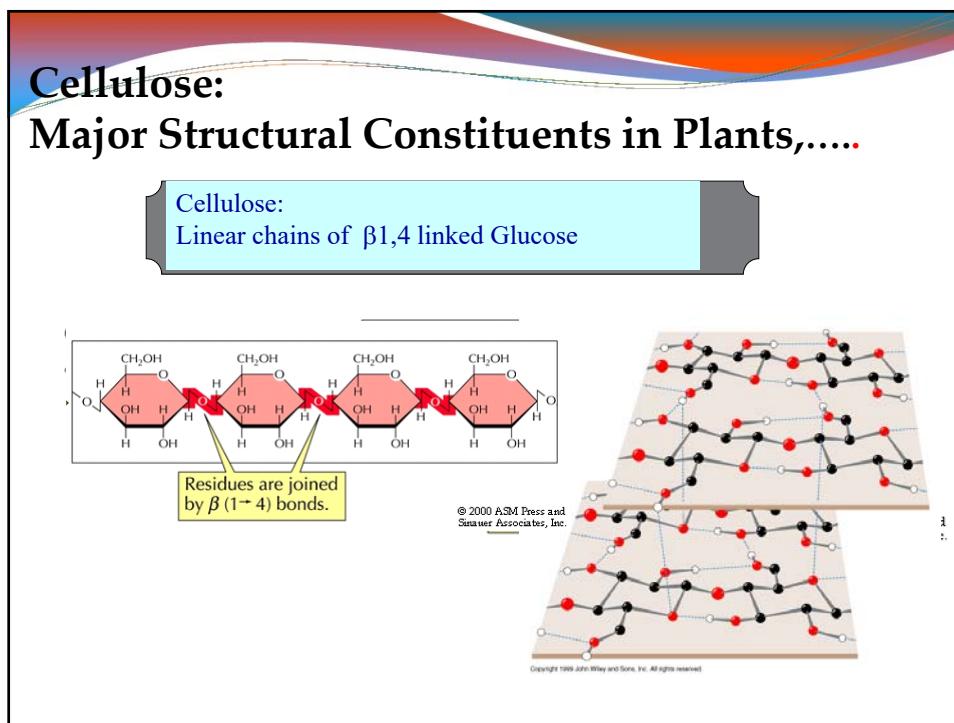
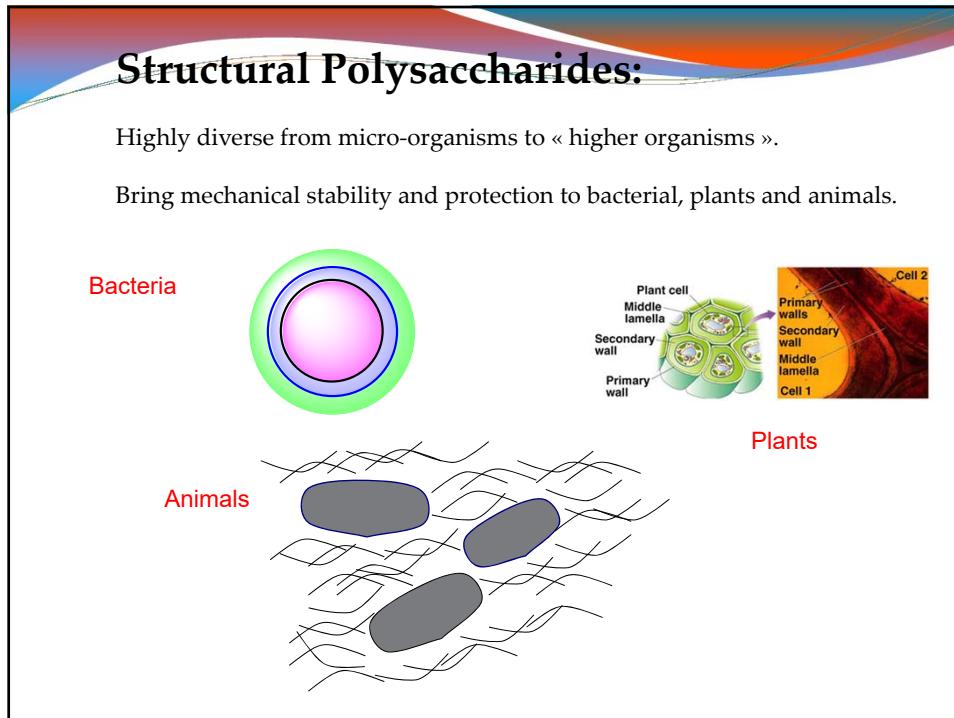
Inulin :

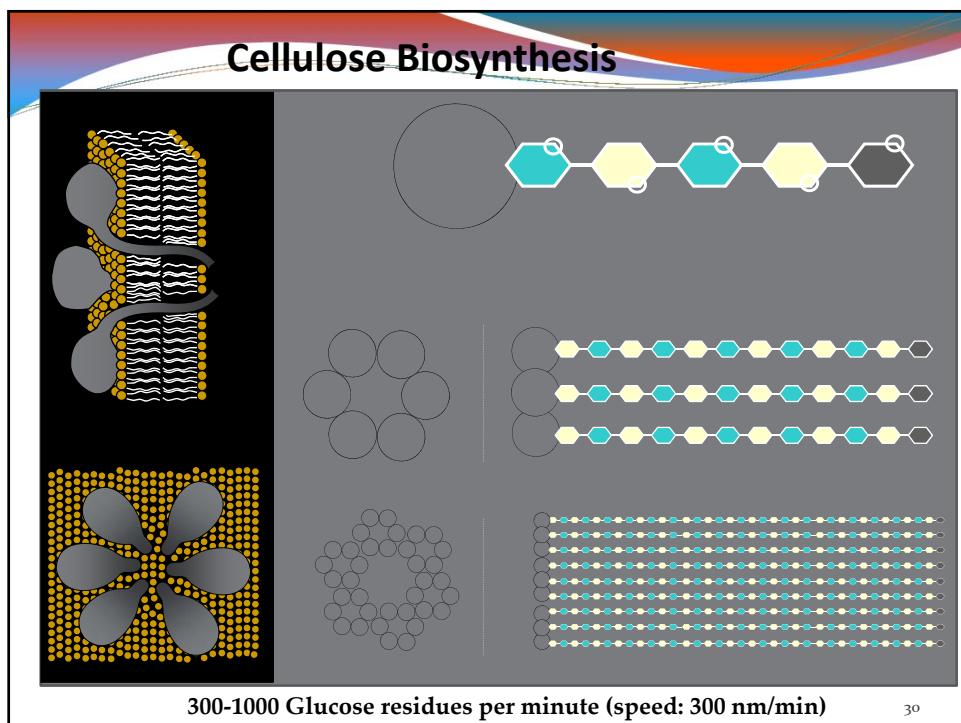


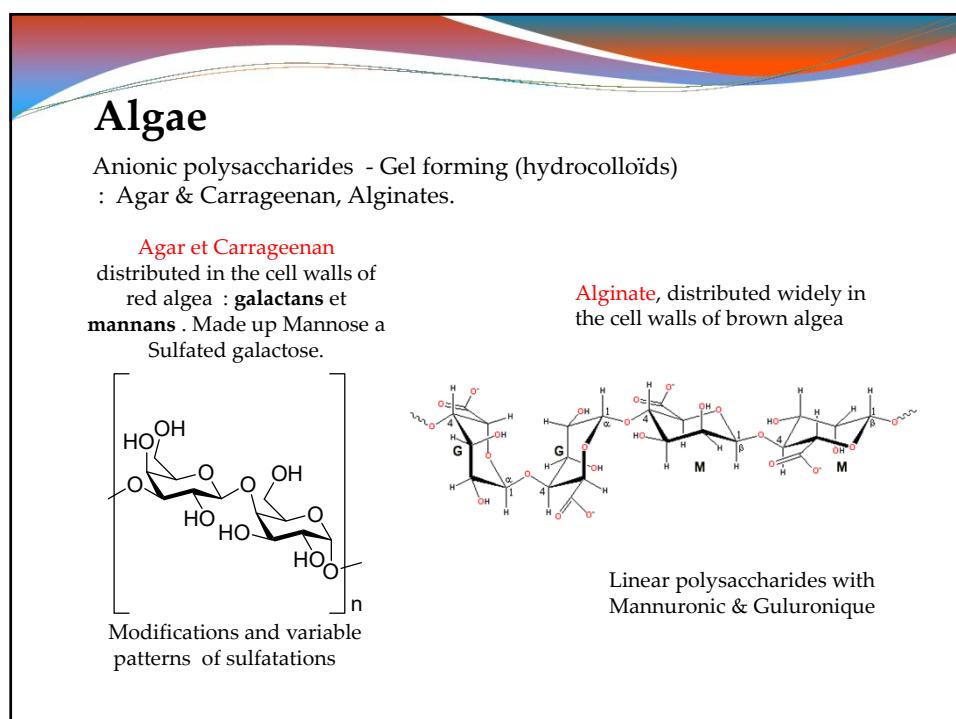
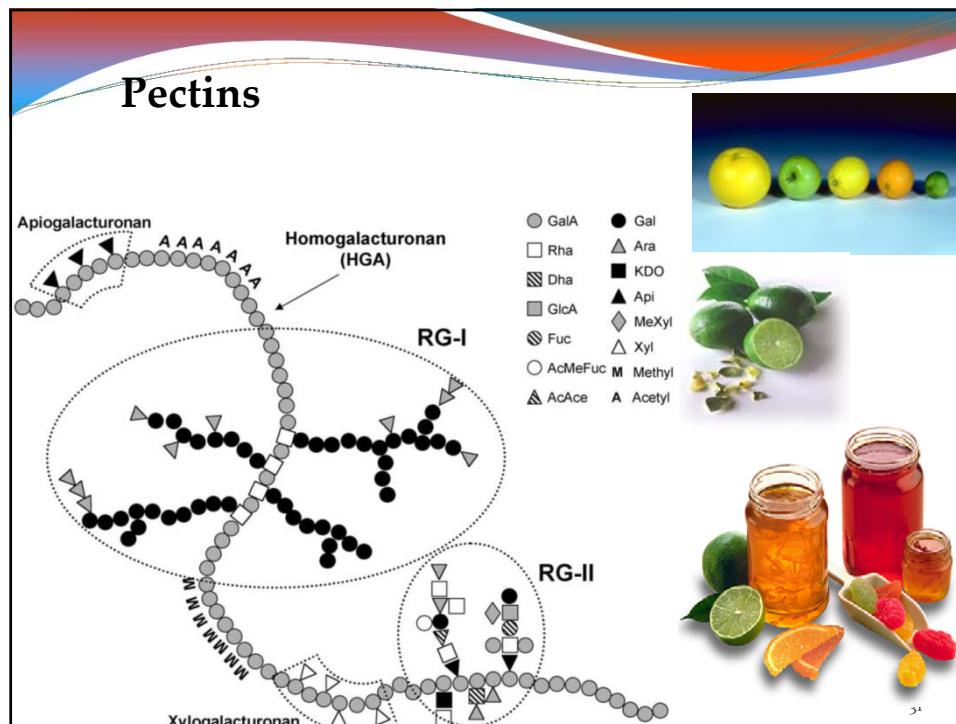
Phlein

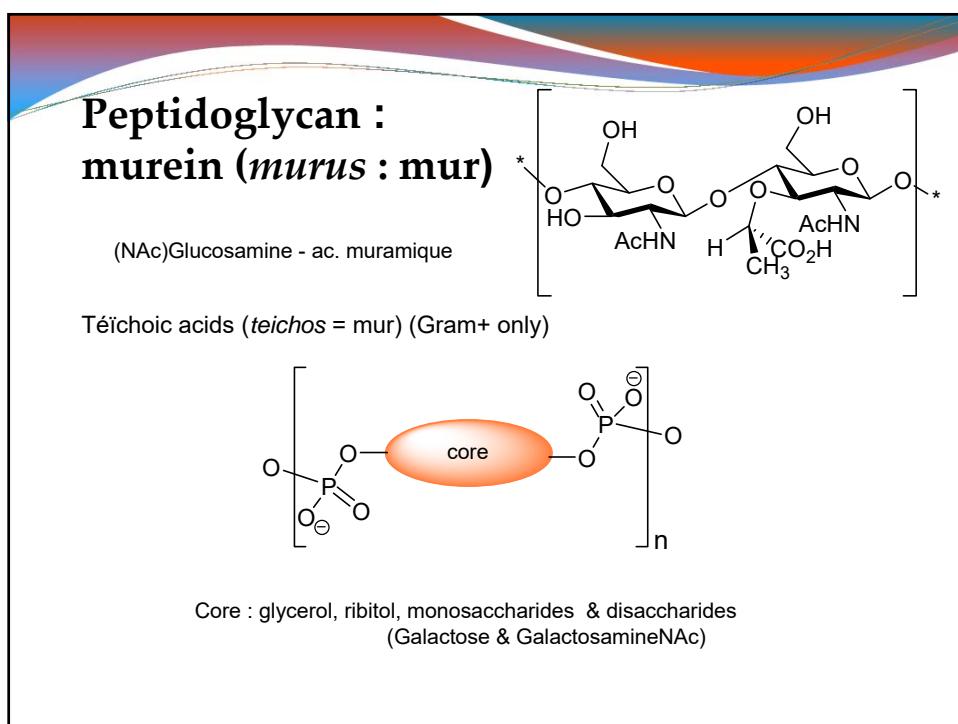
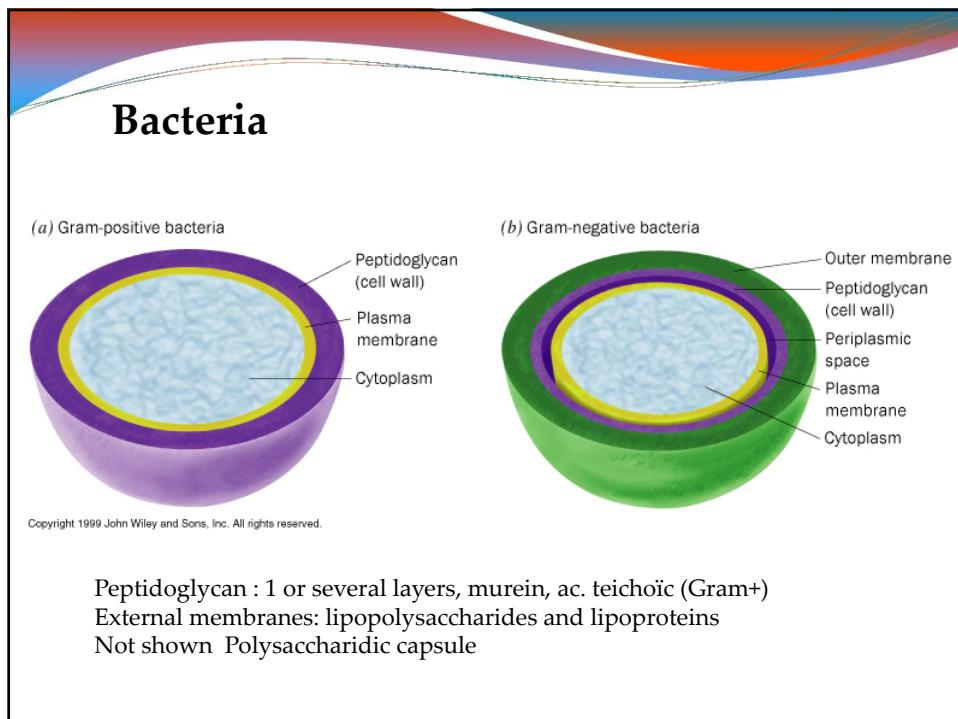


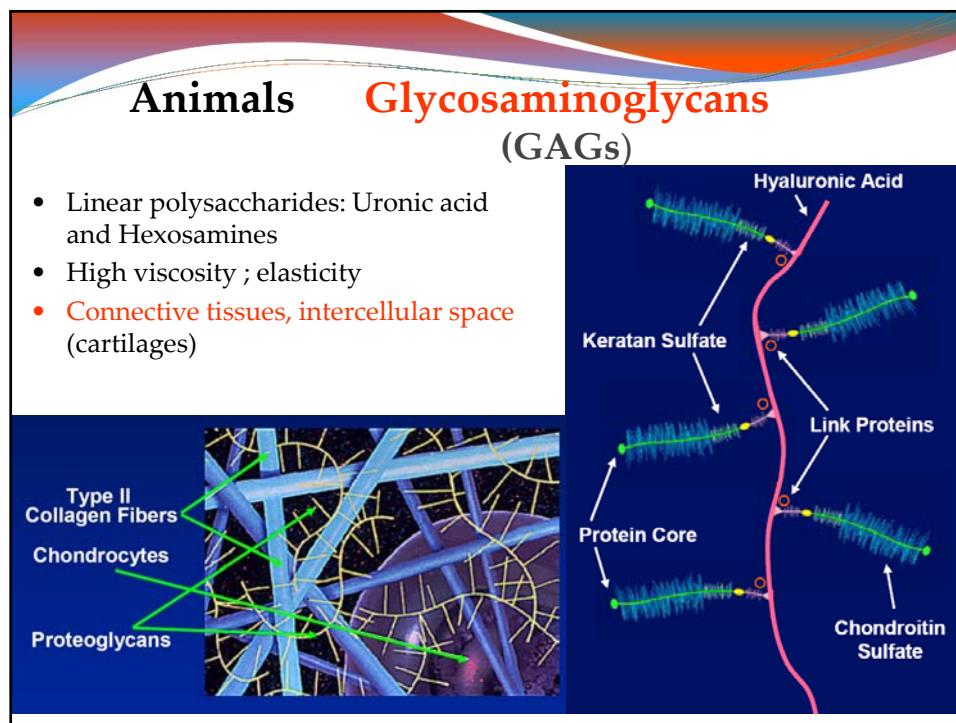
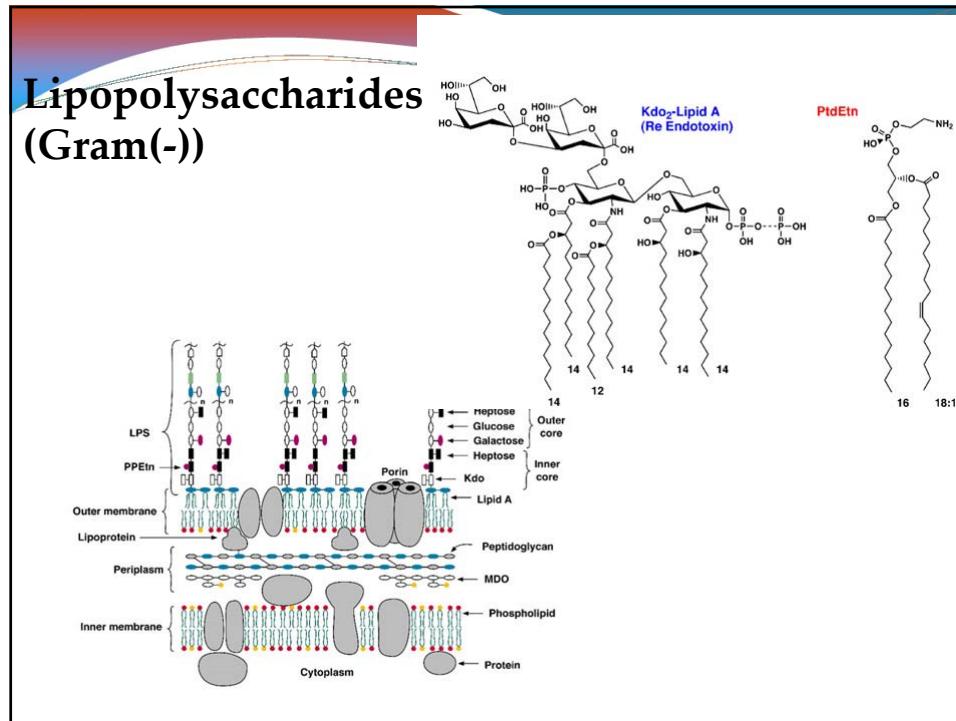
Phlein biosynthesized by phlein sucrase.
High correlation of the chain length of phlein with the stress resistance.

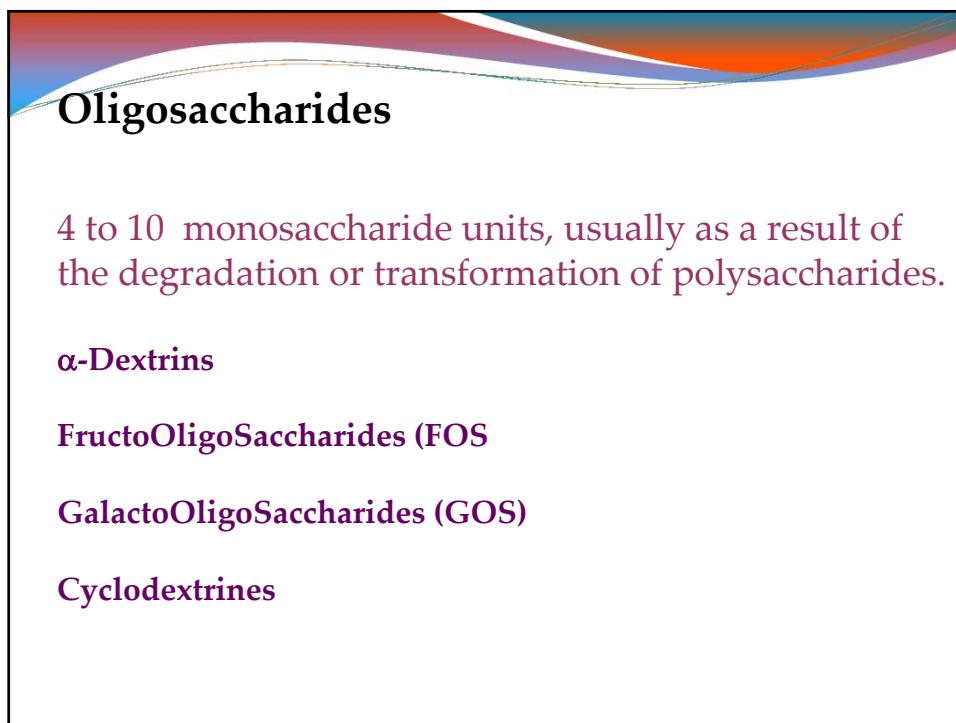
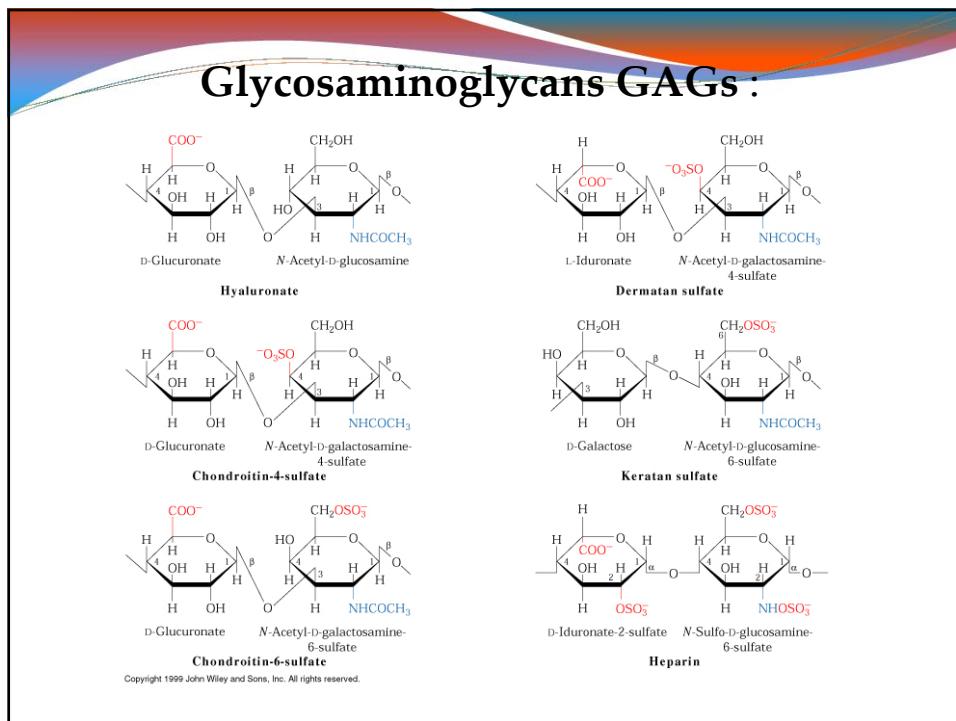












Cyclodextrins : (α , β , γ)
Cyclic Oligomers of α 1-4 linked 6,7,8 Glucoses

Encapsulation
 Enhance stability, solubility, controlled release, masking odor
 Applications: Pharmaceutical, Cosmetics, Food
 Analytical chemistry: separation

The diagram shows a 3D model of a beta-cyclodextrin molecule, which is a cyclic oligomer of glucose units. It also includes a chemical structure of the repeating unit and a photograph of various food items like bread and chips, with a large gray X drawn over it, indicating its use in food products.

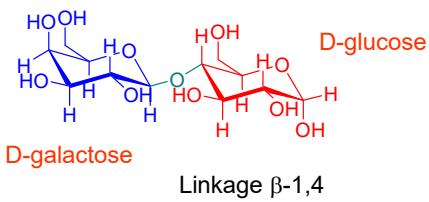
Honey Oligosaccharides
Honey Crystallization

The diagram illustrates the crystallization process of honey. On the left, a bee is shown collecting nectar from a flower. In the center, a jar of honey is being poured. To the right, a molecular model shows the transition from a disordered state (left) to a crystallized state (right), with arrows labeled "Fructose" and "Glucose". Below this, chemical structures of various honey oligosaccharides are shown, including Eriose, Panose, and Thianderose.

F as Fructose & Fat

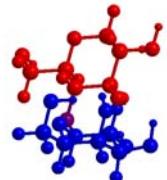
The Most Common Disaccharides

Lactose (galactose – glucose)



D-galactose D-glucose
Linkage β -1,4

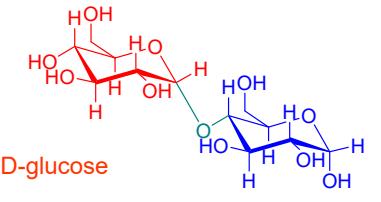





Human Milk Oligosaccharides

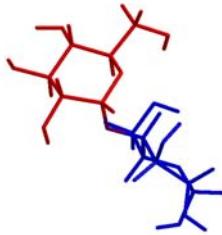
The Most Common Disaccharides

Maltose (glucose-glucose)



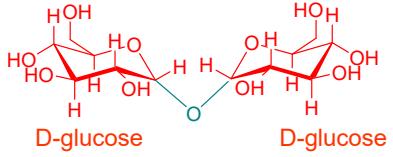
D-glucose D-glucose
Linkage α -1,4





The Most Common Disaccharides

Trehalose (glucose-glucose)



Linkage $\alpha-1,1$



Cryoprotection

Medical Applications :
Stabilization & Protection
antibodies, enzymes, embryos,
help in transplantation
...cosmetic

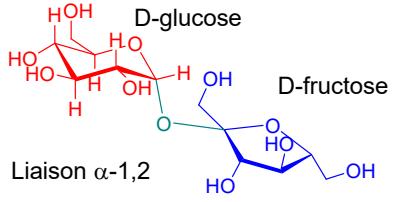
Cryoprotection- Tardigrades



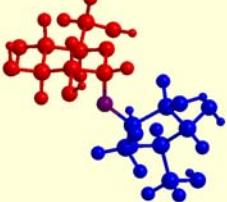
photo by H. Greven, Düsseldorf

The Most Common Disaccharides

Sucrose (glucose-fructose)



Liaison $\alpha-1,2$



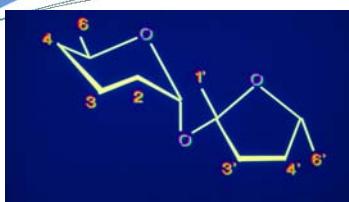



Sugar beat

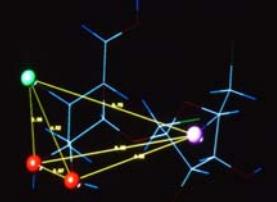


Sugar cane

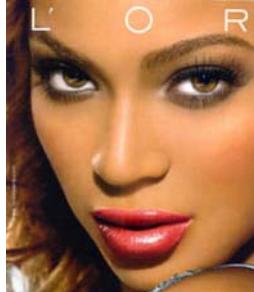
Sucrochemistry.....



	2	3	4	6	1'	3'	4'	6'	Sw
-	-	-	-	-	-	-	-	-	Cl 100
-	-	-	-	-	-	-	-	-	Cl 200
-	-	-	-	-	-	-	-	-	Cl 2200
-	-	-	-	-	-	-	-	-	Cl 200
Cl	-	-	-	-	-	-	-	-	bitter
-	-	-	-	-	-	-	-	-	Br 7500
-	-	-	-	-	-	-	-	-	Cl 400
-	-	-	-	-	-	-	-	-	Cl 500
-	-	-	-	-	-	-	-	-	Cl 0
-	-	-	-	-	-	-	-	-	Cl 1000
-	-	-	-	-	-	-	-	-	Cl 3000
-	-	-	-	-	-	-	-	-	I Cl 3500
-	-	-	-	-	-	-	-	-	Cl Cl 200
-	-	-	-	-	-	-	-	-	Br Br 7000
-	-	-	-	-	-	-	-	-	Cl - F Cl 1000
-	-	-	-	-	-	-	-	-	Cl - Br Cl 3000
-	-	-	-	-	-	-	-	-	I - I Cl 3500
-	-	-	-	-	-	-	-	-	F - Cl Cl 200
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-	-	-	-	-	-	-	-	-	Cl - Cl 4
-	-	-	-	-	-	-	-	-	Cl - Cl 650
-	-	-	-	-	-	-	-	-	Cl - Cl 220
-	-	-	-	-	-	-	-	-	Cl - Cl 160
-	-	-	-	-	-	-	-	-	F - F 40
-	-	-	-	-	-	-	-	-	Br - Br 800
-	-	-	-	-	-	-	-	-	I - I 120




Du rêve à la réalité

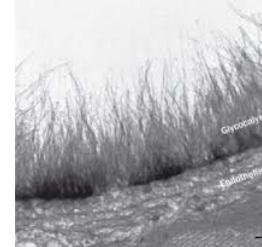


L' O R

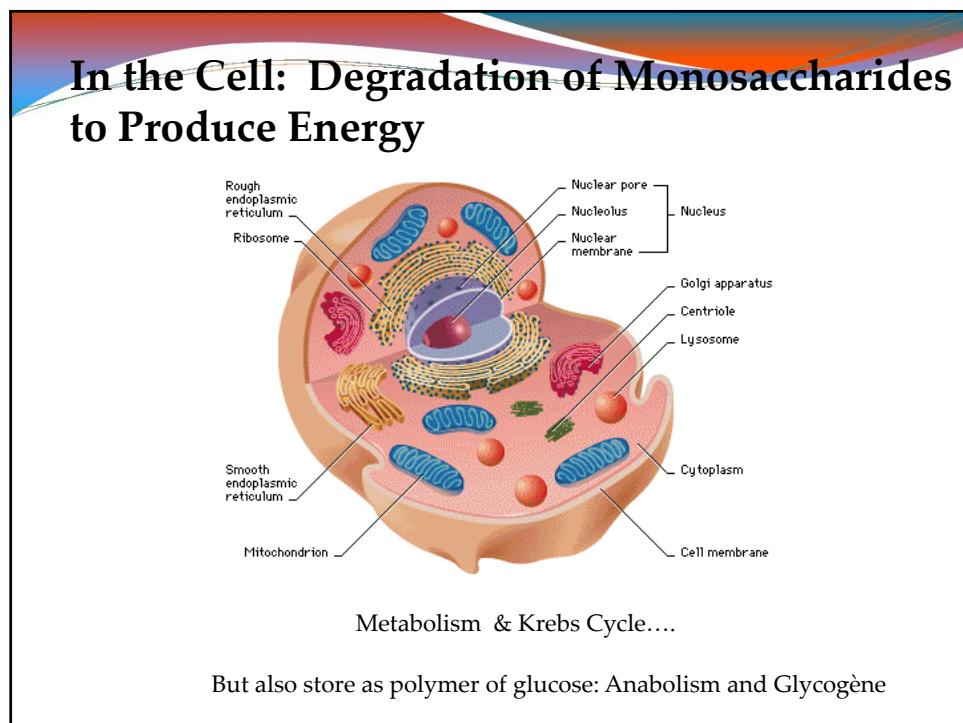
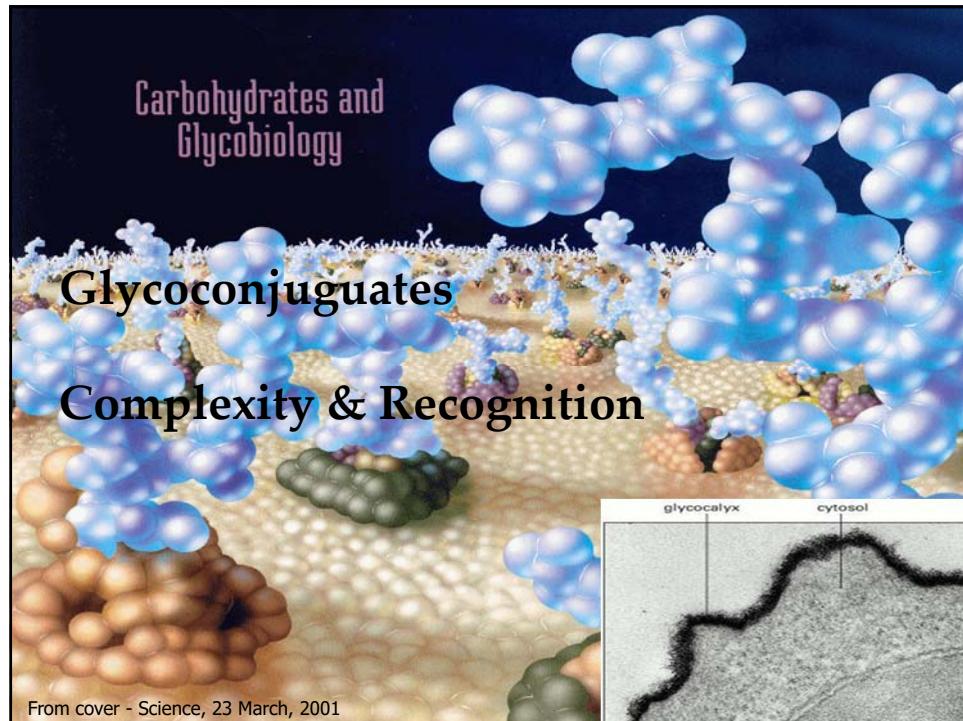


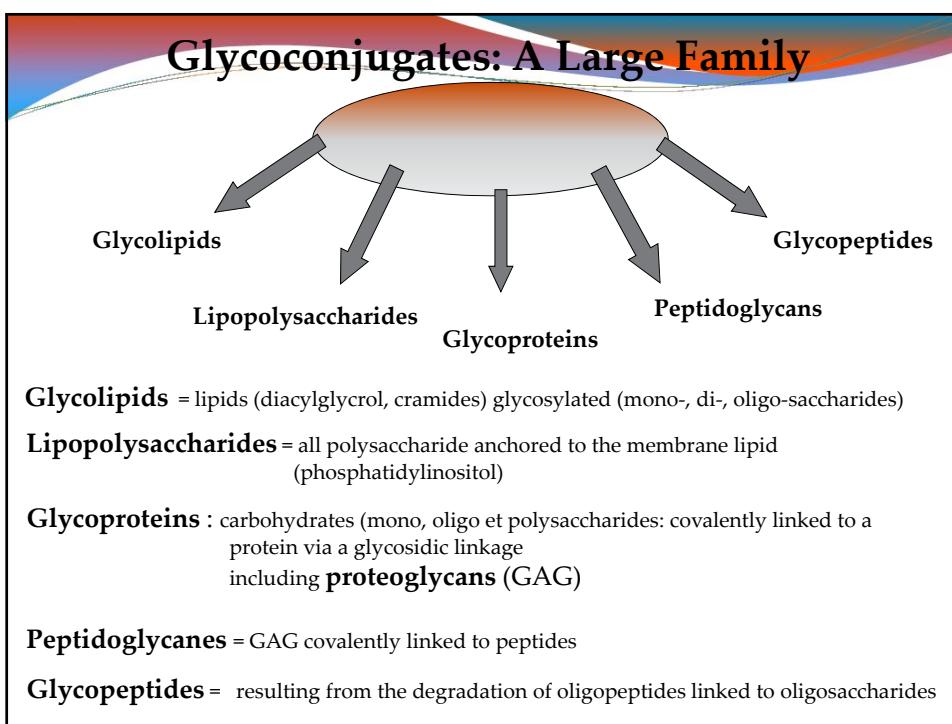
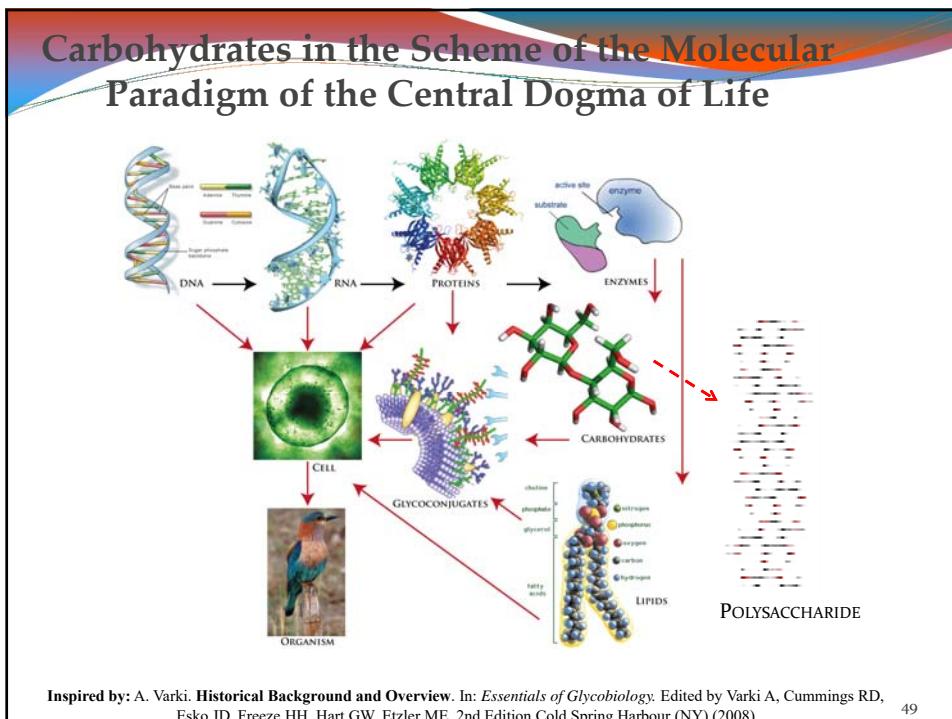


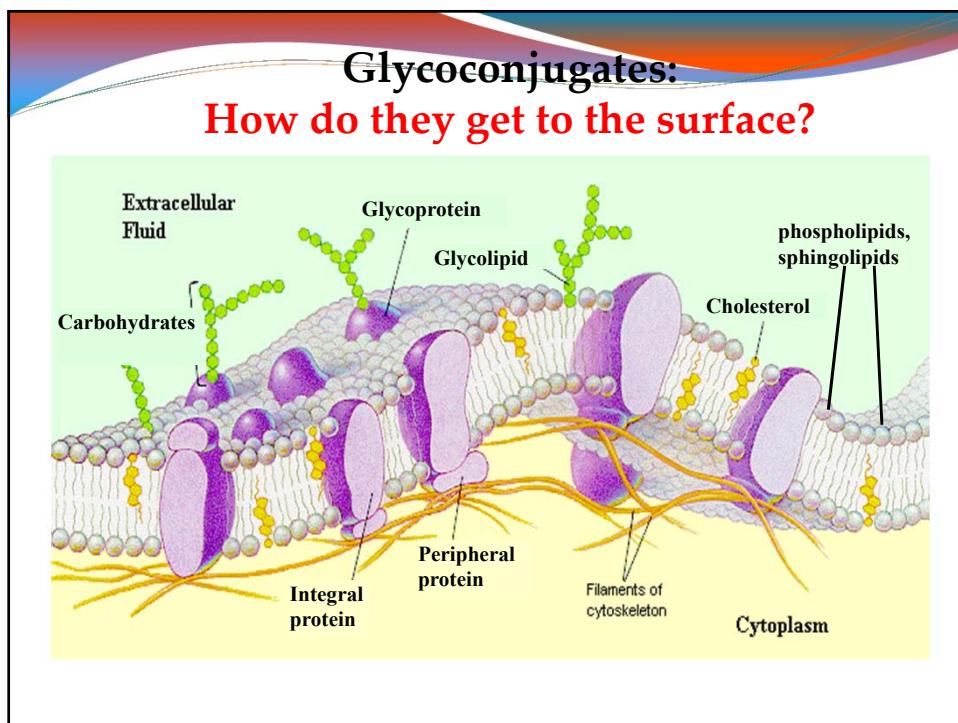
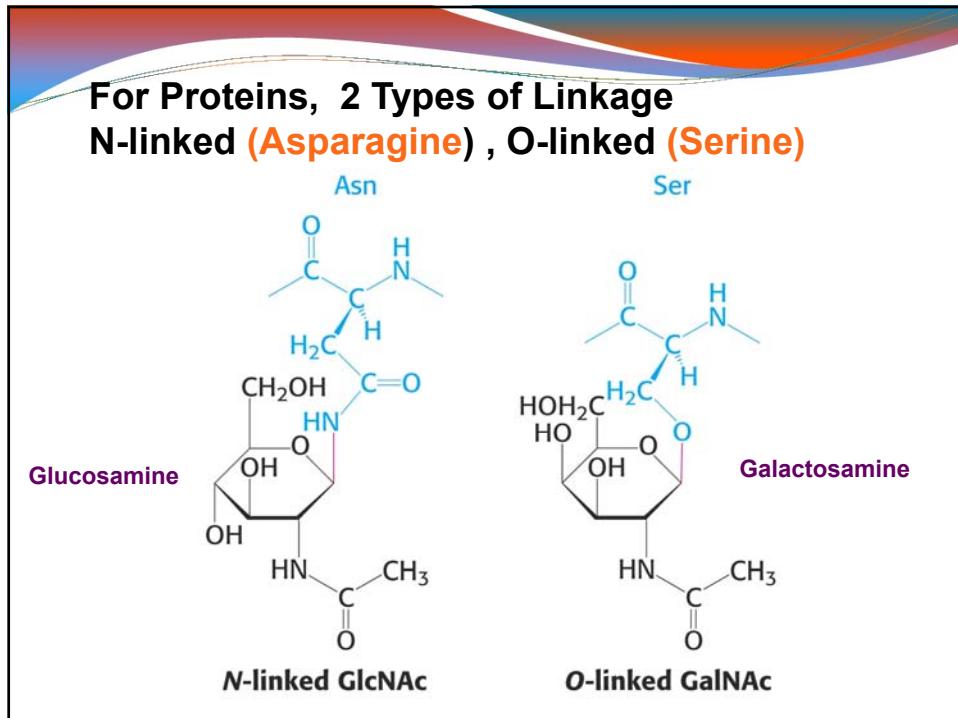
© Science Picture Co

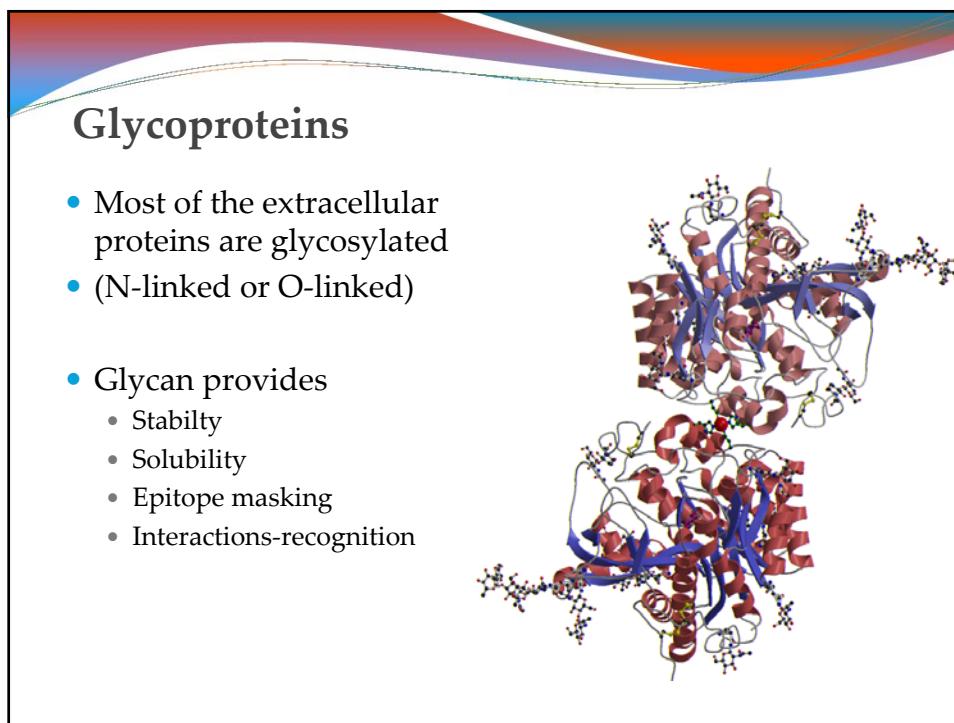
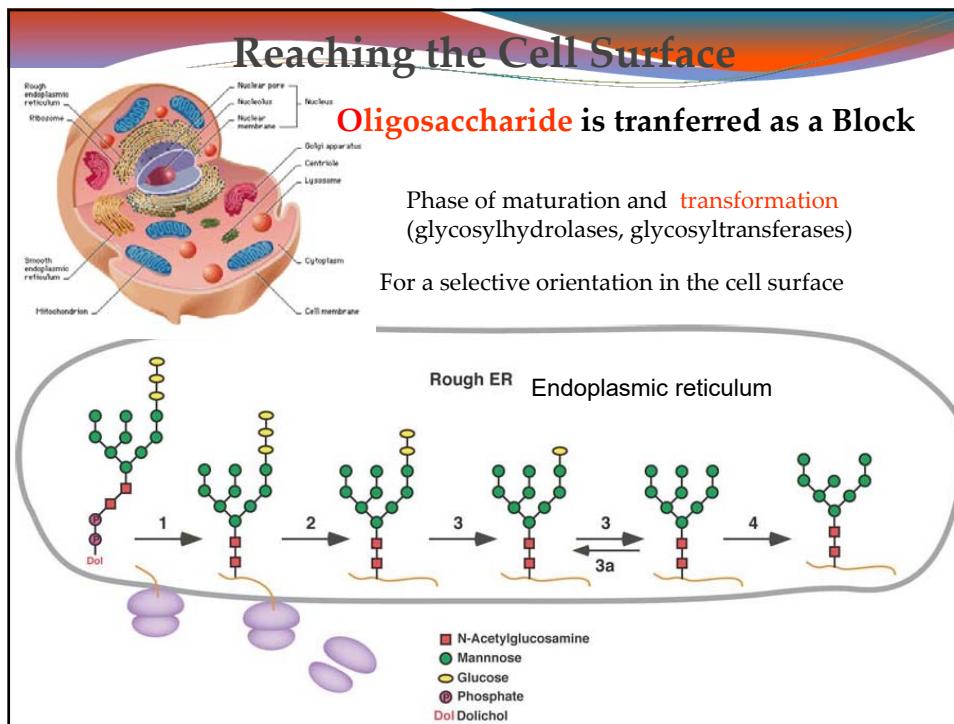


Glycocalyx
Endothelial









Oligosaccharides and Glycoconjugates

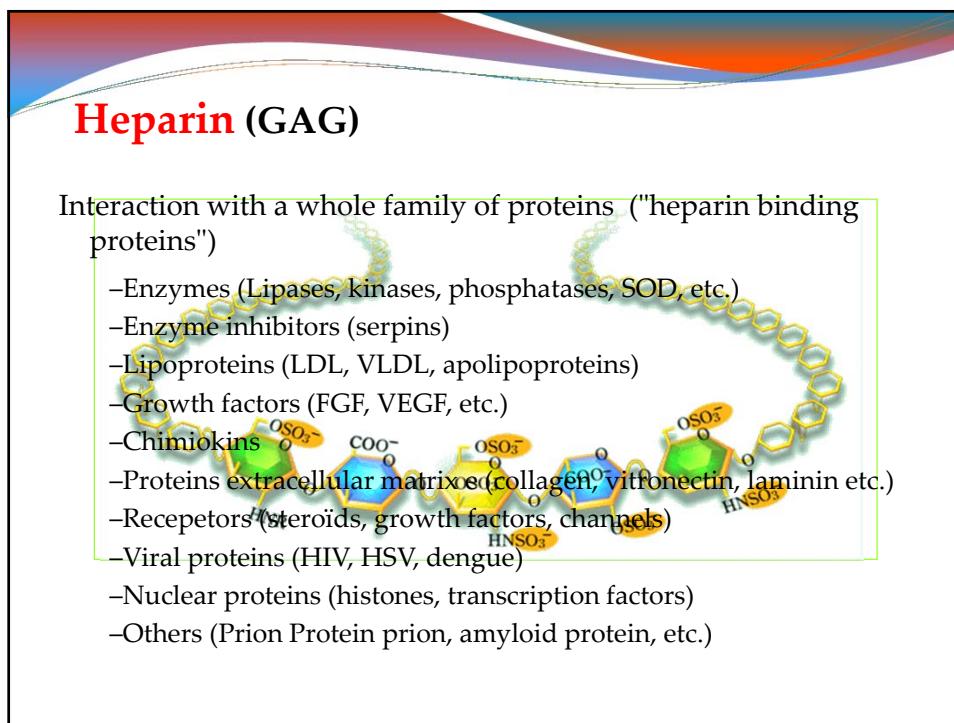
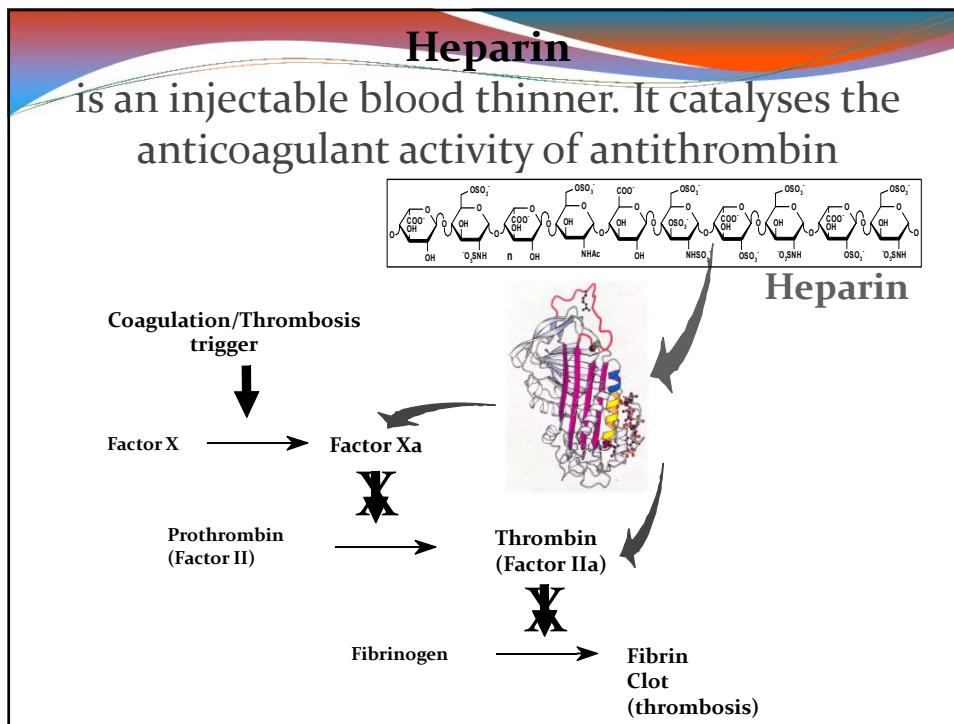
In living cells, glycans can be analogized to identity badges containing access codes to cells and/or cellular organelles.

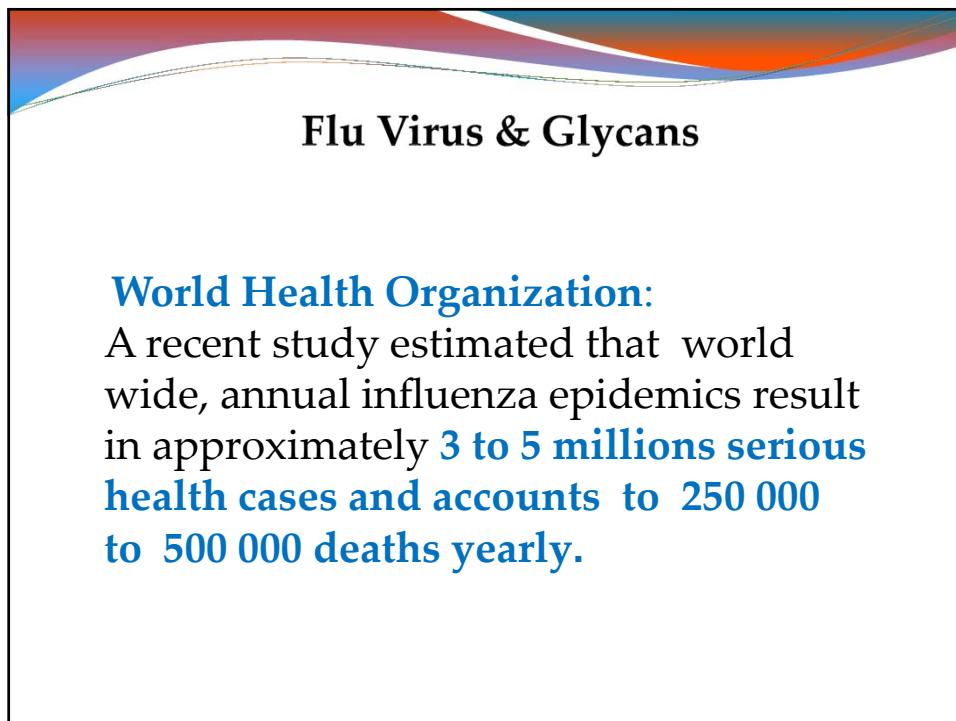
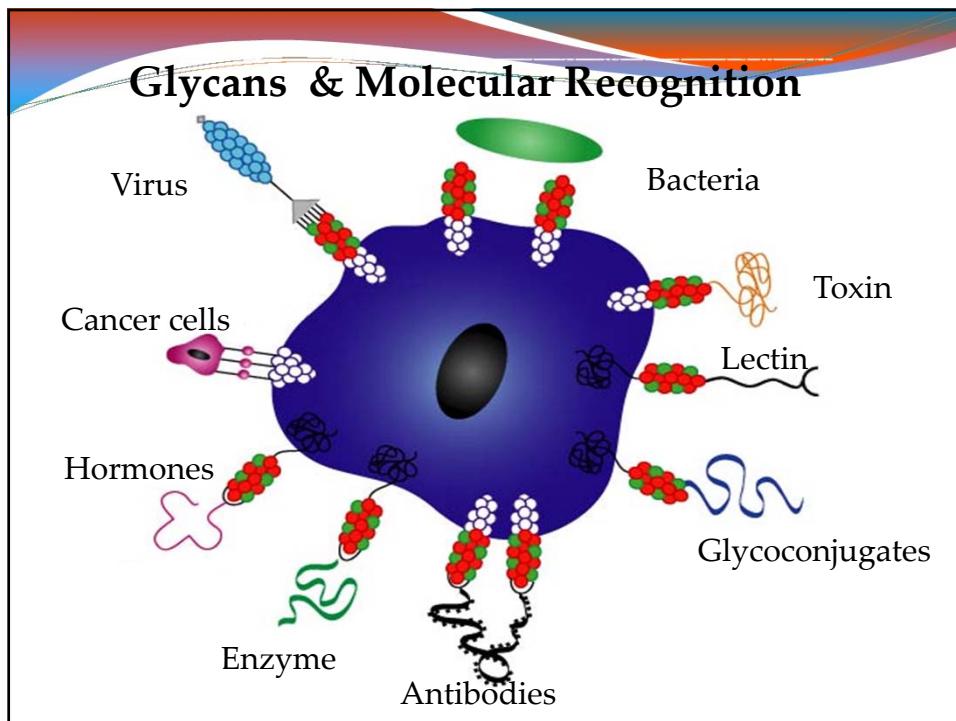
55

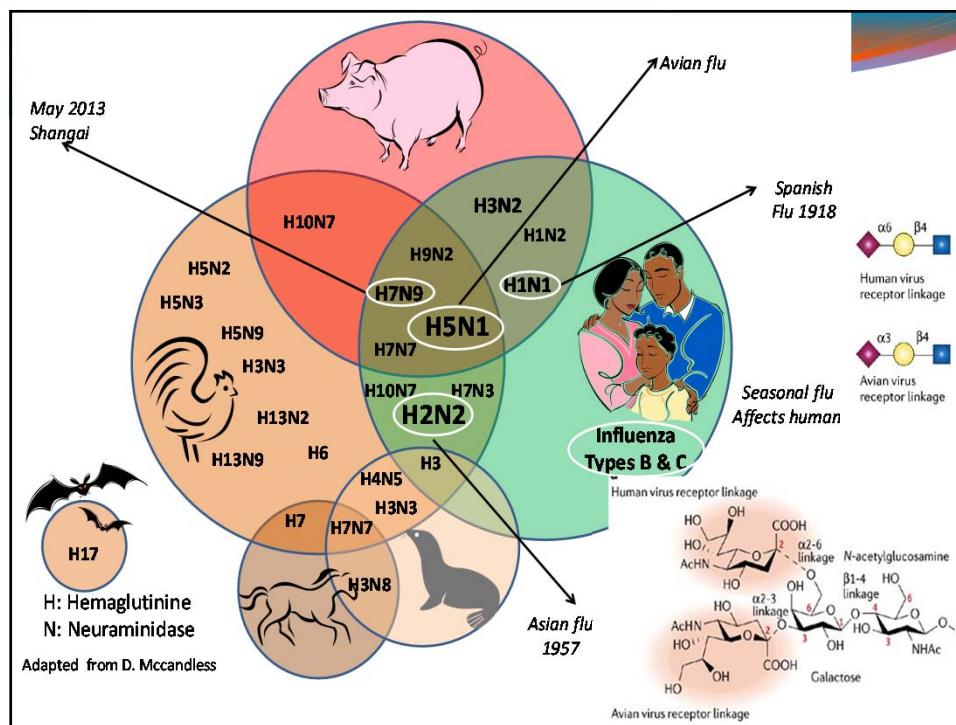
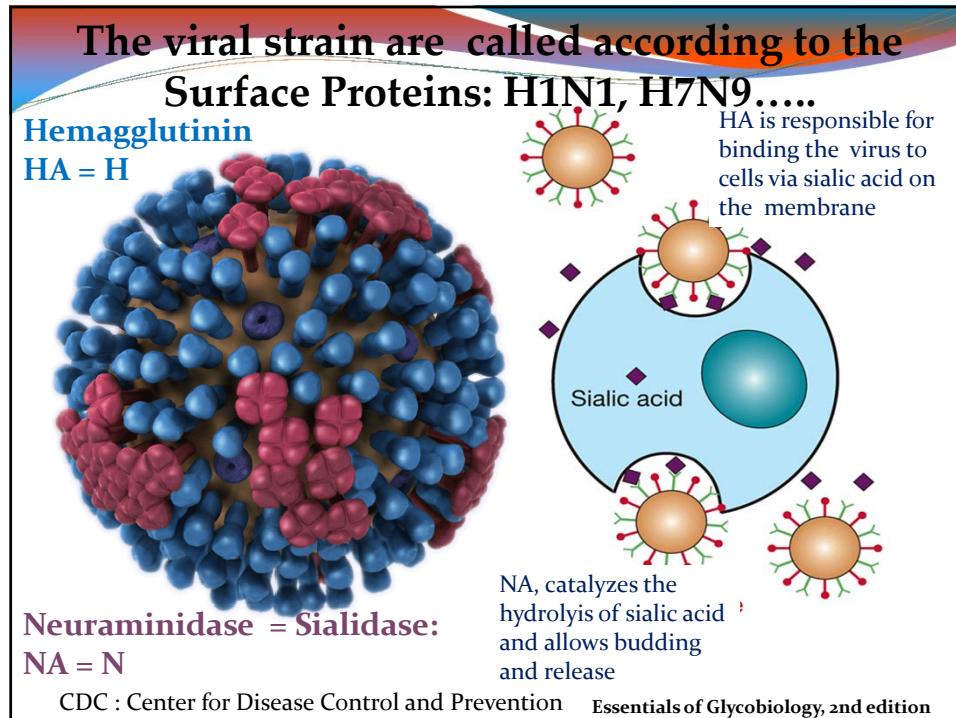
The ProteoGlycans

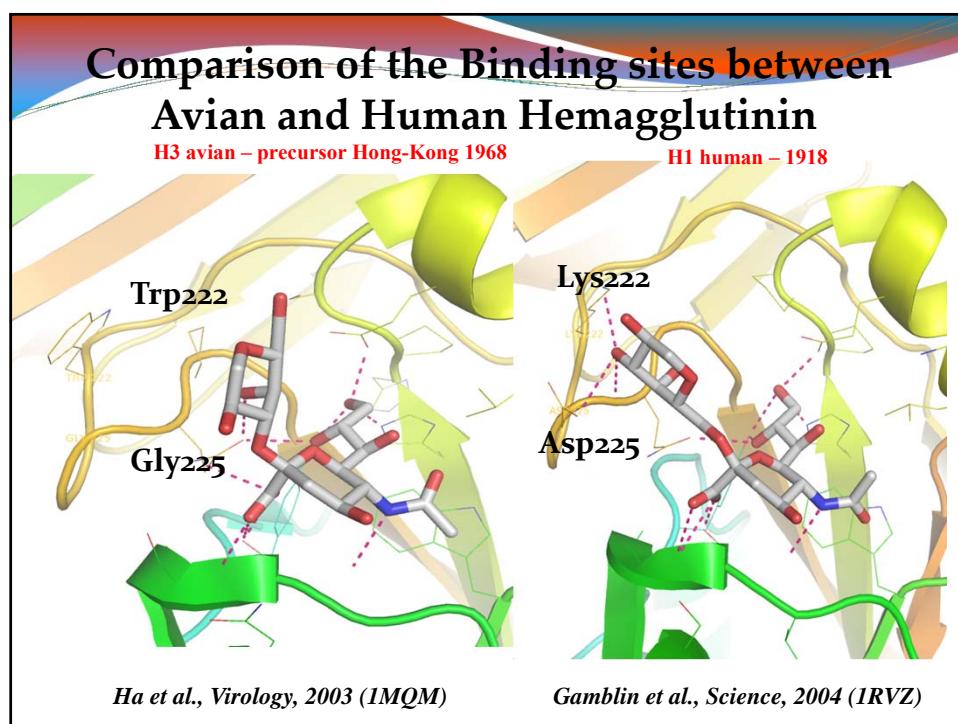
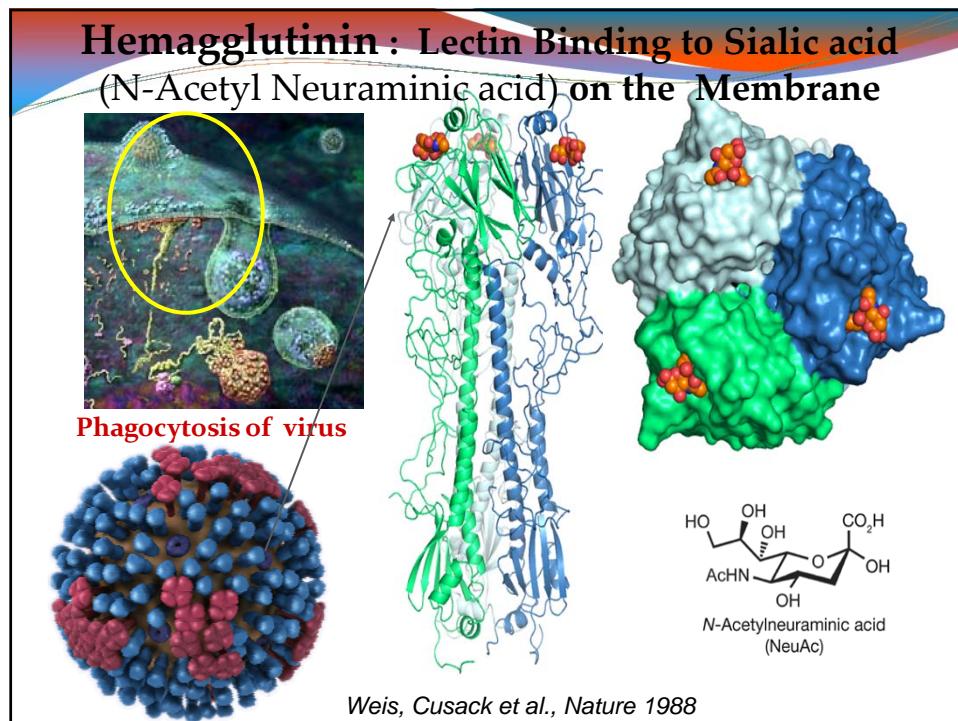
“Core protein” with one or more covalently attached glycosaminoglycans (GAGs) :
occur in the connective tissue Extracellular matrix.

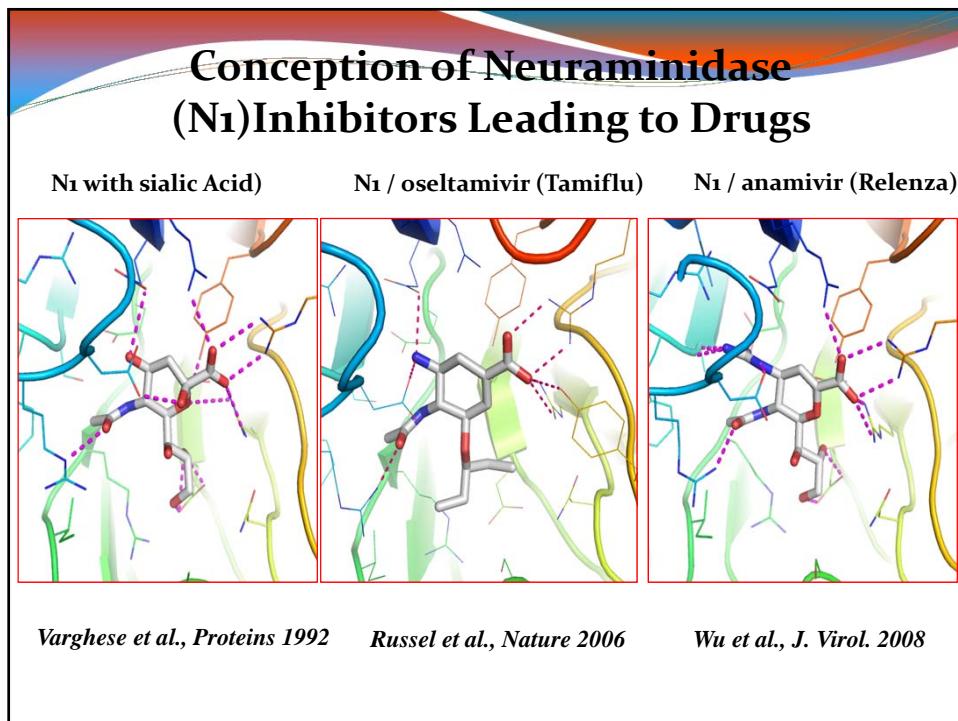
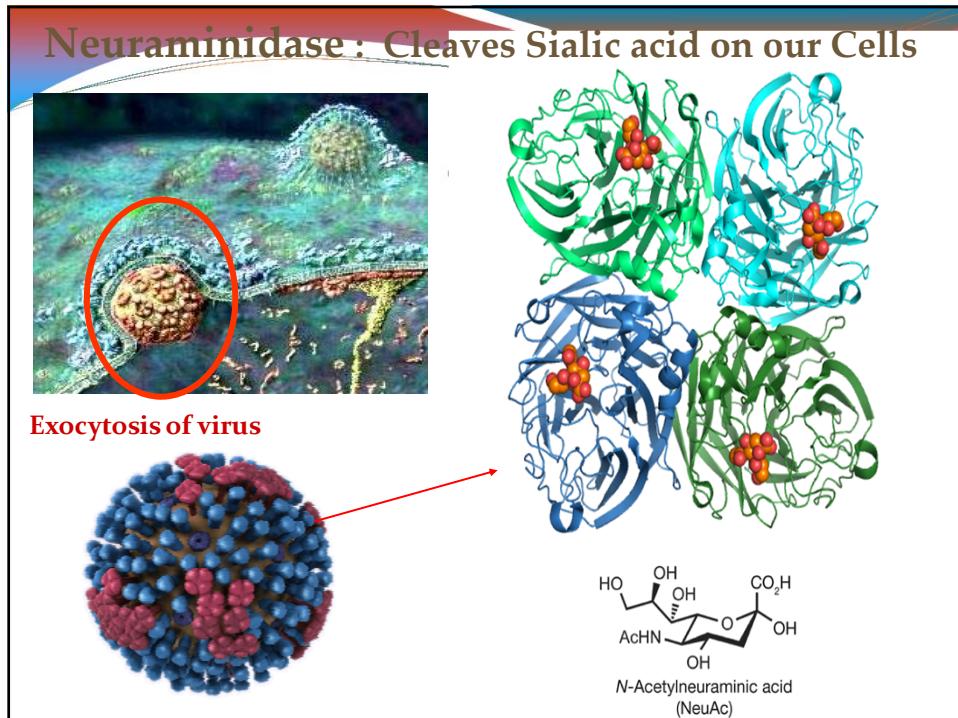
- Chondroitin sulfate
- Keratan sulfate
- Dermatn sulfate
- Hyaluronic acid
- Heparin/Heparan sulfate





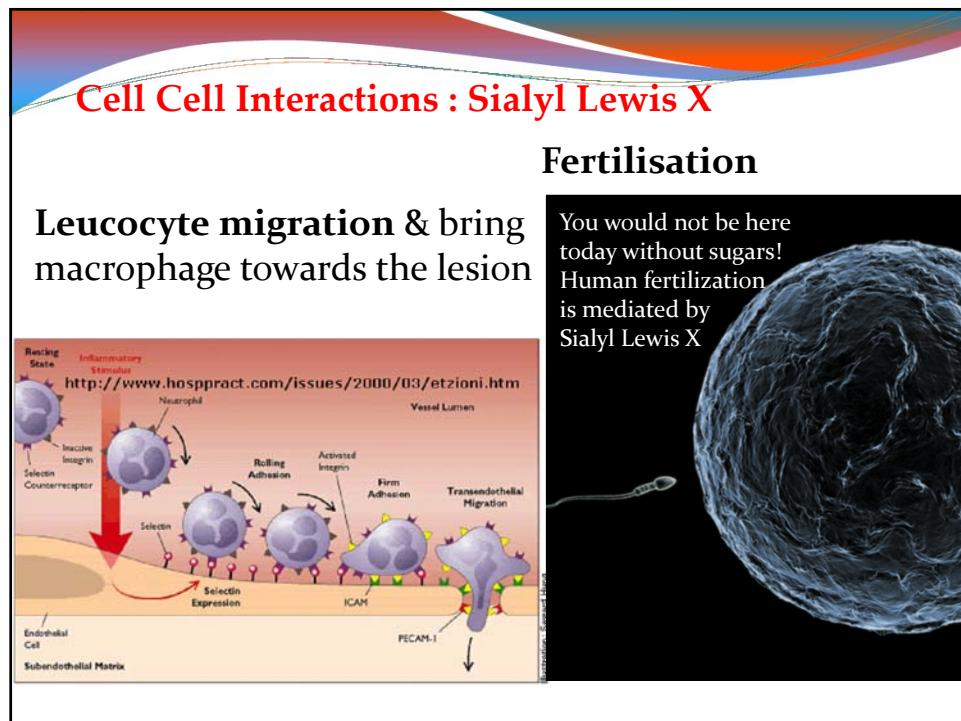


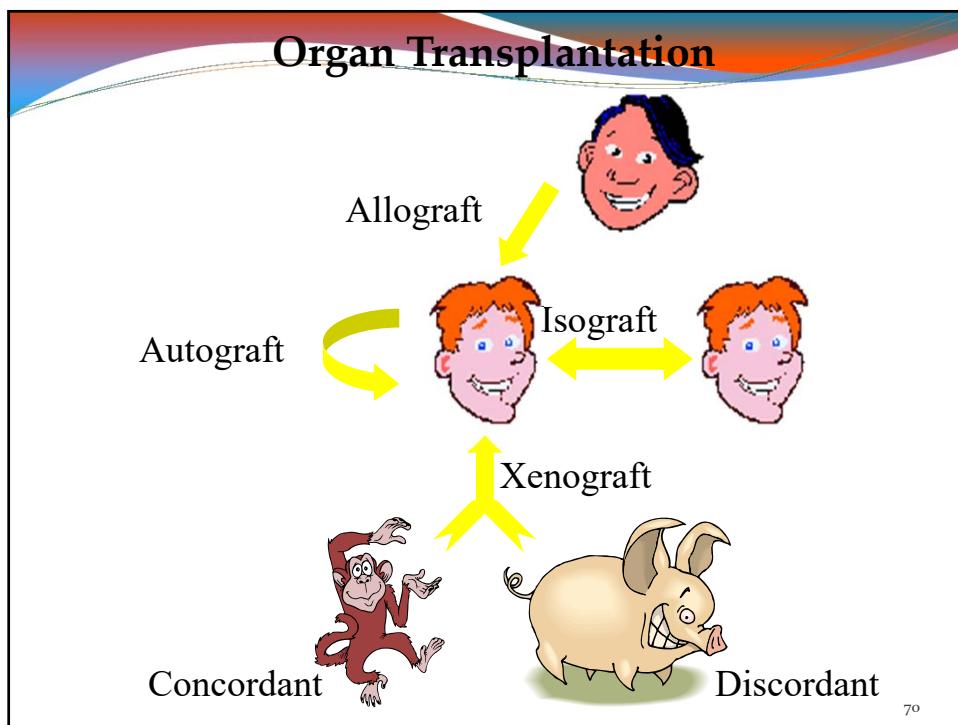
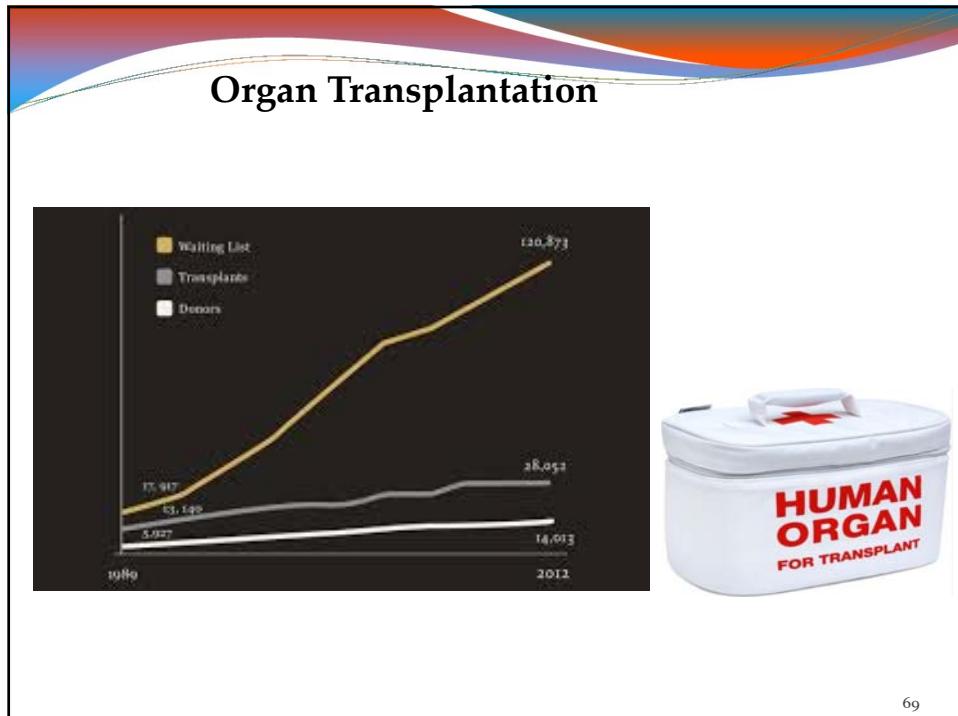


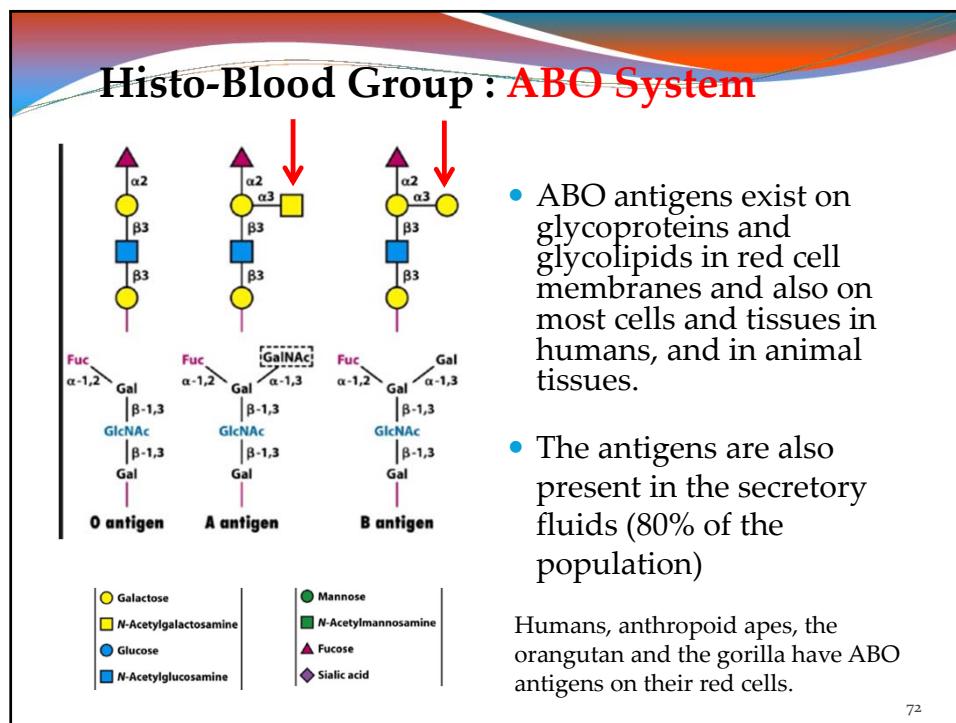
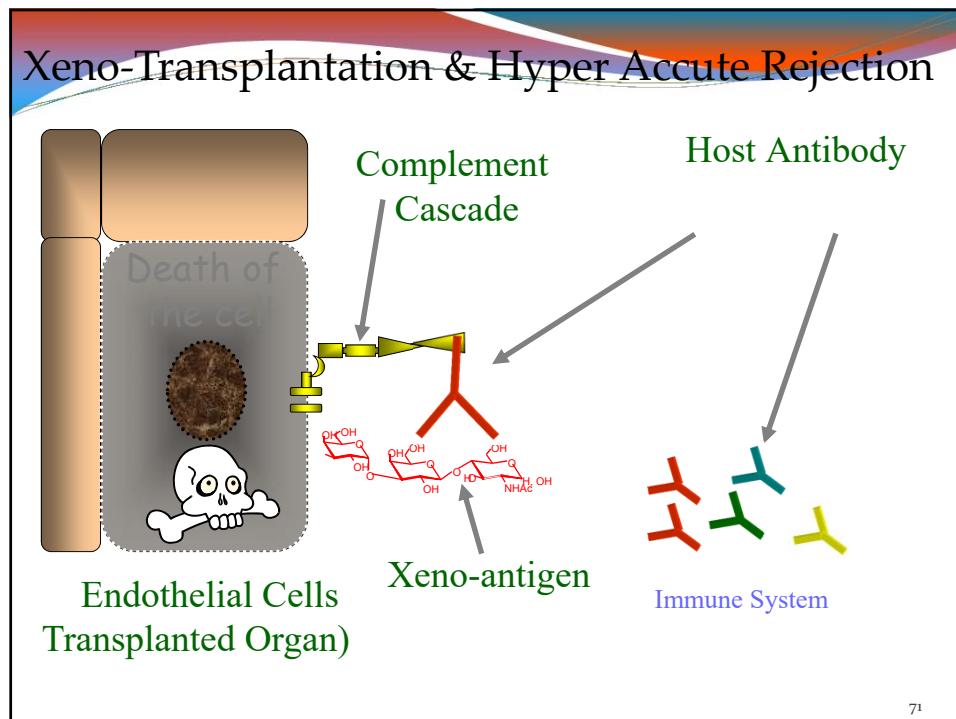


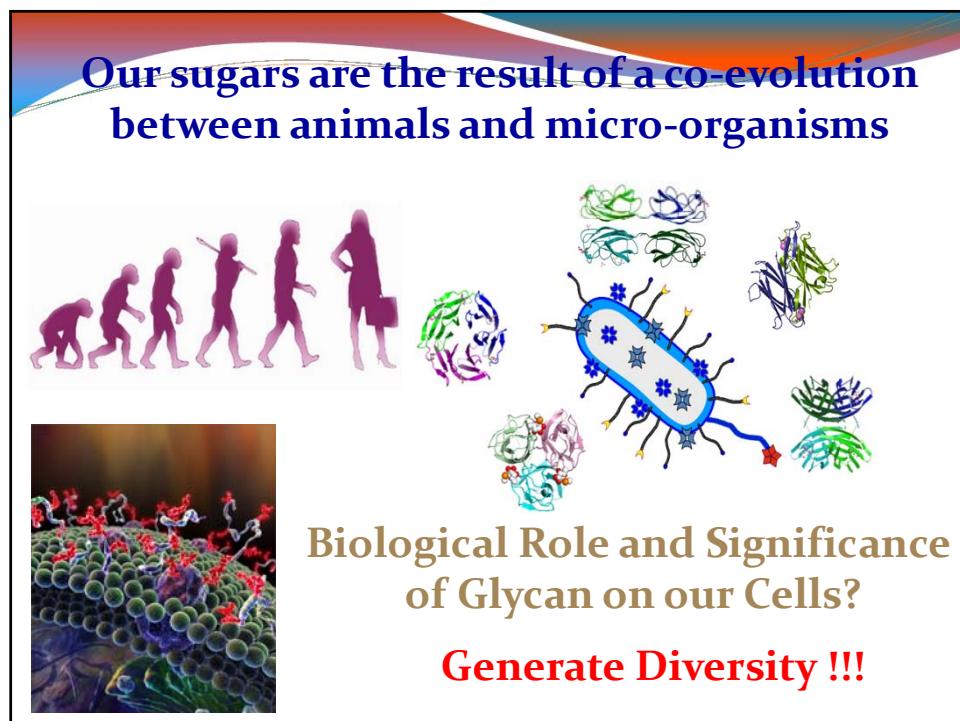
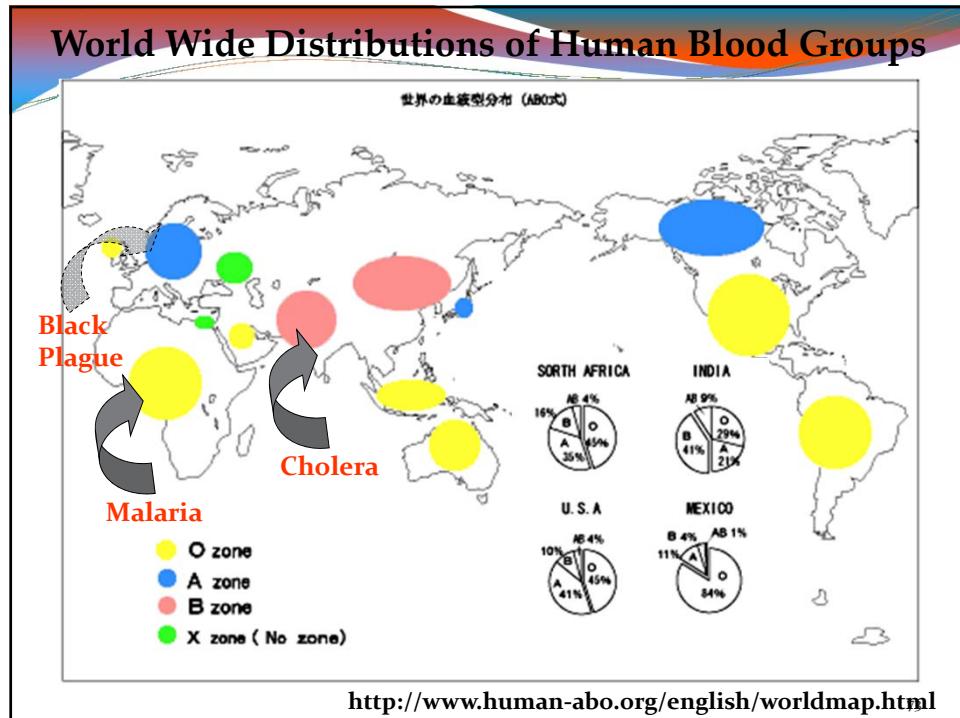
Oligosaccharides as Antigen Determinants	
Tn: Antigen GalNac-Ser	Tumor cells markers
Tf Antigen Gal β 1-3 GalNac – Ser	Tumor cells markers
STn Antigen Neu5Ac α 2-6 GalNAc – Ser	Tumor cells markers (breast cancer)
Sialyl Lewis X Antigen Neu5Ac α 2-3 Gal β 1-4 (Fuc α 1-3) GlcNAc	Cellular Adhesion Fertilisation
Galili Antigen Gal α 1-3 Gal	Hyper acute rejection
Blood Group Antigen : ABO	

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The screenshot shows the homepage of the GlycoPedia website. At the top left is the GlycoPedia logo, which consists of a stylized eye-like shape with a blue outline and a grey interior containing a black dot. To the right of the logo is the URL <http://www.glycopedia.eu/>. Below the URL is a horizontal navigation bar with four tabs: "news", "e-chapters", "resources", and "search". The "resources" tab is highlighted with a blue background. The main content area contains several cards. The first card on the left shows a 3D molecular model and the title "Sweet Unity Mol : 3D Visualization of Complex...". The second card on the right is titled "Online Databases & Tools" and features a grid of colored icons. The third card on the left shows a small icon and the title "A Picture Dictionary of Monosaccharide...". The fourth card on the right shows a photo of a workshop and the title "Workshop Structural Glycoscience. Grenoble...". The fifth card on the left shows a small icon and the title "Library of Bio-active Monosaccharides. 1D,...". The sixth card on the right shows a photo of a city skyline and the title "Glyco Course Firenze 2016". In the bottom right corner of the page, there is a small number "75".